p-PYRIDOYLAMINOBENZOIC ACIDS AND THEIR DERIVATIVES AS POSSIBLE TUBERCULOSTATS *†

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ABSTRACT

p-Pyridoylaminobenzoic acids, their glycine and DL-methionine conjugates and their esters, hydrazides and aldehyde derivatives of the hydrazides totalling 105 compounds have been synthesised and screened for in vitro activity against Mycrobacterium tuberculosis $H_{37}R_v$ strain. Out of these, 19 compounds have shown inhibition of growth of the organism at dilutions varying fiom 1 in 10000 (1007/ml.) to 1 in 10000000 (0.1 $\gamma/ml.$). Three compounds which were highly active in vitro (0.1 γ to 1 $\gamma/ml.$) were found to be ineffective, in vivo in experimental tuberculosis of mice tested by the drug-diet method.

Although the existing potent antituberculosis agents could be used successfully, there are many problems still to be faced to render the treatment more effective. While the development of bacterial resistance, which is a major problem, can be prevented or atleast delayed by the simultaneous or alternating combination of drugs, there is need for newer drugs which, used singly, can eliminate the tubercle bacilli rupidly and radically.

In a search for new antituberculosis agents, N-acetyldipeptides and their derivatives¹, pyridoylaminoacids and their derivatives², pyridylacrylic acids, pyridylpropionic acids, furylacrylic acids and their derivatives³, chlorophenoxy-acetic acids and their derivatives⁴ and pyridoyldipeptides and their derivatives⁵ have been synthesised in our laboratory and screened for *in vitro* antituberculosis activity with encouraging results. The present work consists of the synthesis and testing of the following types of compounds.

^{*}Abstracted from the Ph.D. thesis, Indian Institute of Science, Bangalore, 1966, of Sri B. K. Mohan Murali.

[†] Presented at the Seminar on Recent Advances in Pharmaceutical Sciences held at Chandigarh in September, 1966 under the auspices of University Grants Commission (India).

- (A) p-Isonicotinoylaminobenzoic acid, its glycine and DL-methionine conjugates, their esters, hydrazides and hydrazone derivatives (39 compounds).
- (B) p-Nicotinoylaminobenzoic acid, its glycine and DL-methionine conjugates, their esters, hydrazides and hydrazone derivatives (39 compounds).
- (C) p-Picolinoylaminobenzoic acid, its glycine and DL-methionine conjugates, their esters, hydrazides and hydrazone derivatives (27 compounds).

The scheme of synthesis given in Chart I is self-explanatory.

While isonicotinic acid⁶ and picolinic acid⁷ were prepared by the permanganate oxidation of γ - and α -picolines respectively, nicotinic acid was available from stock. The acid chlorides of the pyridine carboxylic acids (II) have been prepared by treating their potassium salts with thionyl chloride in dry benzene⁸. The acid chloride on condensation with ethyl *p*-aninobenzoate gave the corresponding ethyl *p*-pyridoylaminobenzoates (III) in good yields. Ethyl esters of glycine and DL-methionine conjugates (VIII) of *p*-pyridoylaminobenzoic acids have been prepared via the acid azide method⁹, *i.e.*, by the action of ethyl ester of the aminoacid on the acid azides of *p*-pyridoylaminobenzoic acids.

Screening for *in vitro* tuberculostatic activity was carried out by the serial dilution method using surface culture technique by Professor M. Sirsi. The substances were first dissolved in ethylene glycol and further required dilutions were prepared in the Youman's media. The test organism was *Mycobacterium tuberculosis* $H_{37}R_{\gamma}$ strain. The degree of inhibition or growth was observed at weekly intervals. The degree of inhibition or growth seen at the end of three weeks was recorded. The results of *in vitro* screening are mentioned in the experimental part and discussed at the end.

EXPERIMENTAL*

(A) p-ISONICOTINOYLAMINOBENZOIC ACID, ITS GLYCINE AND DL-METHIONINE CONJUGATES AND THEIR DERIVATIVES

Ethyl p-isonicotinoylaminobenzoate¹⁰ (Compound No. 1)

To a suspension of potassium isonicotinate (20 g.) in dry benzene (100 ml.), cooled to 0° C, was slowly added redistilled thionyl chloride (40 ml) over a period of 10 min. After addition, the reaction mixture was refluxed for 1 hr. The crude isonicotinoyl chloride obtained on removal of benzene and excess of thionyl chloride was used as such for further condensation.

^{*}All melting points are uncorrected and d denotes decomposition. Infrared spectra of all the compounds taken in the Perkin-Eimer Infracord Model 137B snowed the characteristic absorptions.



To a stirred solution of this acid chloride in dry benzene (300 ml.) was added a solution of ethyl *p*-aminobenzoate (20.5 g.) dissolved in dry benzene (200 ml.) during 1 hr. at room temperature. After the mixture was kept stirred for 6 hr., water (200 ml.) was added, the mixture neutralized with sodium bicarbonate and kept stirred for another hour. As the product was msoluble in benzene, it was filtered, washed with benzene and water, dried and crystallised from aqueous ethanol m.p. 154-5°C. (Reported¹⁰ m.p. 140°C). Yield: 30 g.; 91%. (Found: 66 18; H, 5.09; N, 10.79. $C_{15}H_{14}O_3N_2$ requires: C, 66.66; H, 5.22; N, 10.36%)

p-Isonicotinoylaminobenzoic acid¹⁰ (Compound No. 2)

Ethyl *p*-isonicotinoylaminobenzoate (l g.) was stirred with potassium hydroxide solution (IN, 20 ml.) for 3 to 4 hr. at room temperature. The precipitated acid was filtered, washed, dried and crystallised from ethanol. m p. 350° C.(d). (Reported¹⁰ m p 350° C.). Yield · 0.4 g.: 45%.

p-Isonicotinoylaminobenzoic acid hydrazide¹⁰ (Compound No. 3)

A mixture of ethyl *p*-isonicotinoylaminobenzoate (20 g.) dissolved in ethanol (400 ml, 90%) and hydrazine hydrate (22 g., 99%) was refluxed for 6 to 8 hr. Alcohol and excess of hydrazine hydrate were distilled off under suction and the solid residue obtained was crystalised from ethyl alcohol. m.p. 246-8°C. (Reported¹⁰ m.p. 235-7°C.). Yield: 12 g.: 63%. (Found: C, 60.60; H, 490; N, 21 69. C_{13} H₁₂O₂N₄ requires: C, 60 93; H, 4.72; N, 21.86%).

Aldehyde derivatives of p-isonicotinoylaminobenzoic acid hydrazide

These were prepared by refluxing in each case, *p*-isonicotinoylaminobenzoic acid hydrazide (1 g.) and an equimolar amount of the aldehyde in absolute ethanol (100 ml.) for δ hr. The hydrazones which precipitated on cooling were filtered, washed with a little alcohol, dried and crystallised from ethanol. Table 1 contains the compounds prepared (Nos. 4 to 13) in this series.

Ethyl p-isonicotinovlaminobenzoylglycinate (Compound No. 14)

To a stirred solution of *p*-isonicotinoylaminobenzoic acid hydrazide (12.8 g.) in dilute hydrochloric acid (90 ml., 1:1), maintained at 0° C., was added a previously cooled solution of sodrum nitrite (25 g.) in water (60 ml.) over a period of 30 min. After 2 hr. of stirring, the mixture was neutralized with sodium carbonate solution and the solid azide filtered, washed and drued over phosphorus pentoxide in a vacuum desiccator. Yield: 12.6 g.; 94%.

To a stirred solution of *p*-isonicotinoylaminobenzoic acid azide (12.3 g.) in dimethylformamide (DMF) (100 ml.) was added ethyl glycinate (7.11 g.) dissolved in DMF (10 ml.) over a period of 30 min. at rcon temperature. After stirring for 10 hr., the solution was filtered and the solvent partly removed under suction. Addition of either benzene-petrol mixture (1:1) or water gave the required product which was filtered and crystallised from ethanol. m p. 207-8°C. Yield: 12.6 g.; 83.6%. (Found: C, 62.46; H, 5.54; N, 12.92 C_{17} H₁₇ O₄ N₃ requires : C, 62.38; H, 523; N, 12.84%).

	Ň	>CONH	SCON	H-N=CH	∃−R	
	<u> </u>	.//			R=Alde	hye residue
Con	nd Aldehyde	Mol. formula	m.p.°C	Fou	nd/Required	1
No	. residue			c	H	N
4	C ₆ H ₅ -	$C_{20}H_{16}O_2N_4$	292 - 3(d)	69 31	4.77	16.41
				69.76	4 68	16 27
5	o-OH C ₆ H₄-	C ₂₀ H ₁₆ O ₃ N ₄	289 - 90(d)	66.36	4.28	15.69
				66.66	4.48	15.55
6	<i>m</i> -OH C ₆ H₄-	C ₂₀ H ₁₆ O ₃ N ₄	310 (d)	66.90	4,88	15.49
				66.66	4 48	15.55
7	<i>p</i> -OH.C ₆ H ₄ -	$C_{20}H_{16}O_{3}N_{4}$	334 - 5(d)	66.22	4 83	15.22
				66 66	4 48	15.55
8	C₅H₄N-3-	$C_{19}H_{15}O_2N_5$	285 (d)	65.80	3 99	19.89
				66.08	4 38	20.28
9	C₅H₄N-4-	C ₁₉ H ₁₅ O ₂ N ₅	280 - 1(d)	66.33	4.33	19.80
				66.08	4 38	20,28
10	p-MeO C ₆ H ₄ -	C ₂₁ H ₁₈ O ₃ N ₄	290 (d)	67 30	4.83	14 62
				67.37	4.85	14.96
11	p-Me2N.C6H4-	C22H21O2N5	294 (d)	67.72	5.50	18.28
				68 20	5.46	18 08
12	$C_6H_5CH = CH-$	$C_{22}H_{18}O_2N_4$	291 (d)	71.25	4.96	15.21
				71.34	4.90	15.13
13	C4H30-2-	C13H14O3N4	318 (d)	64.28	4 38	16.37
				64.67	4.22	16.76

Tabi.f	1
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Aldehyde Derivatives of p-Isonicotinoylaminobenzoic Acid Hydrazide

Note: d means decomposition

p-Isonicotinoylaminobenzoylglycine (Compound No. 15)

This was prepared by the hydrolysis of ethyl *p*-isonicotinoylaminobenzoylglycinate (0.5 g.) with potassium hydroxide solution (1N, 10 ml.) and isolated in the usual manner. m.p. 298°C.(d). Yield: 0.2 g.; 43.8%. (Found: C, 59.74; H, 4.07; N, 13.68. $C_{15} H_{13} O_4 N_3$ requires: C, 60.20; H, 4.38; N, 14.04%). p-Isonicotinoylaminobenzoylglycine hydrazide (Compound No. 16)

This was prepared by refluxing ethyl p-isonicotinoylaminobenzoylglyci. nate (8 g.) and hydrazine hydrate (7.3 g.; 99%) in ethanol (400 ml., 90%) in the usual way. m.p. above 300°C. (d). Yield : 7.1 g. ; 92.7%. (Found : C, 57.07; H, 5.12; N, 22.08. C15 H 15 O3 N5 requires : C, 57.50; H, 4.83; N, 22.35%).

Aldehyde derivatives of p-isonicotinoylaminobenzoylglycine hydrazide

The hydrazones were prepared in good yields in the usual way. Table 2 contains the compounds prepared (Nos. 17 to 26) in this series.

$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	Com-	R-	Mal formula		Four	nd/Require	d
17 $C_6H_5^{\bullet}$ $C_{22}H_{19}O_3N_5$ $262-3$ 65.40 4.45 18 $o-OH.C_6H_4^{-}$ $C_{22}H_{19}O_4N_5$ 268 63.54 4.31 19 $m-OH.C_6H_4^{-}$ $C_{22}H_{19}O_4N_5$ 280 (d) 63.17 4.14 19 $m-OH.C_6H_4^{-}$ $C_{22}H_{19}O_4N_5$ 280 (d) 63.17 4.14 63.30 4.59 20 $p-OH.C_6H_4^{-}$ $C_{22}H_{19}O_4N_5$ $310-11$ (d) 62.94 4.27 21 $C_5H_4N-3 C_{21}H_{18}O_3N_6$ 282 (d) 62.38 4.52 22 $C_5H_4N-4 C_{21}H_{18}O_3N_6$ 288 (d) 62.30 4.40 23 $p-MeO.C_6H_4^{-}$ $C_{23}H_{21}O_4N_5$ $268-9$ (d) 64.16 4.50 24 $p-Me_2N.C_6H_4^{-}$ $C_{24}H_{24}O_3N_6$ 281 (d) 64.91 509 64.85 5.44 $506-1$ (d) 67.51 4.50 25 $C_6H_5-CH = CH C_{24}H_{24}O_3N_5$ $260-1$ (d) 67.51 4.50 26 $C_{14}0^22_{-}$ $C_{14}H_{21}O_3N_5$ $260-1$ (d) 67.51 4.50 26 $C_{14}0^22_{-}$ $C_{14}H_{21}O_3N_5$ $260-1$ (d) 67.51 4.50	No.	residue	Mot. formula	ш.р. С	С	(%) H	N
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	17	C ₆ H ₅ -	$C_{22}H_{19}O_3N_5$	262-3	65.40 65.83	4.45 4.77	17 14 17.45
19 m -OH C_6H_4 - $C_{22}H_{19}O_4N_5$ 280 (d)63.174.1463.304.5920 p -OH. C_6H_4 - $C_{22}H_{19}O_4N_5$ 310-11 (d)62.944.2763.304.5921 C_5H_4N -3- $C_{21}H_{18}O_3N_6$ 282 (d)62.384.5222 C_5H_4N -4- $C_{21}H_{18}O_3N_6$ 288 (d)62.304.4023 p -MeO. C_6H_4 - $C_{23}H_{21}O_4N_5$ 268 -9 (d)64.164.5024 p -Me $_2N.C_6H_4$ - $C_{24}H_{24}O_3N_6$ 281 (d)64 915 0964.855.4425 C_6H_5 -CH = CH- $C_{24}H_{21}O_3N_5$ 260 -1 (d)67.514.5926 $C_{14}0^2$ - $C_{14}H_{10}O_3N_5$ 260 -1 (d)67.514.5064.855.445.445067.444.95	18	ø-OH.C ₆ H₄-	$C_{22}H_{19}O_4N_5$	268	63.54 63.30	4.31 4.59	16.78 16.78
20 p -OH.C ₆ H ₄ - C ₂₂ H ₁₀ O ₄ N ₅ 310-11 (d) 62.94 4.27 63.30 4.59 21 C ₅ H ₄ N-3- C ₂₁ H ₁₈ O ₃ N ₆ 282 (d) 62.38 4.52 62.68 4.51 22 C ₅ H ₄ N-4- C ₂₁ H ₁₈ O ₃ N ₆ 288 (d) 62.30 4.40 23 p -MeO.C ₆ H ₄ - C ₂₃ H ₂₁ O ₄ N ₅ 268-9 (d) 64.16 4.50 64.03 4.91 24 p -Me ₂ N.C ₆ H ₄ - C ₂₄ H ₂₄ O ₃ N ₆ 281 (d) 64 91 5 09 64.85 5.44 25 C ₆ H ₅ -CH = CH- C ₂₄ H ₂₄ O ₃ N ₅ 260-1 (d) 67.51 4.50 67.44 4.95 26 C ₁ H ₁₀ C ₂ C ₁ H ₁₀ O ₁ N ₅ 285 (d) 616 4.37	19	<i>т</i> -ОН С ₆ Н ₄ -	$C_{22}H_{19}O_4N_5$	280 (d)	63.17 63.30	4.14 4.59	17.00 16.78
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	20	<i>p</i> -OH.C ₆ H ₄ -	$C_{22}H_{19}O_4N_5$	310-11 (d)	62.94 63.30	4.27 4.59	17.12 16.78
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	21	C5H4N-3-	$C_{21}H_{18}O_3N_6$	282 (d)	62.38 62.68	4.52 4.51	20.73 20 89
23 $p-MeO.C_6H_4$ - $C_{23}H_{21}O_4N_5$ 268 - 9 (d) 64.16 4.50 64.03 4.91 24 $p-Me_2N.C_6H_4$ - $C_{24}H_{24}O_3N_6$ 281 (d) 64 91 5 09 64.85 5.44 25 C_6H_5 -CH = CH- $C_{24}H_{21}O_3N_5$ 260 - 1 (d) 67.51 4.50 67.44 4.95 26 C_H-0 ⁻² C_{24}H_{21}O_3N_5 285 (d) 61 60 4 37	22	C5H4N-4-	$C_{21}H_{18}O_{3}N_{6}$	288 (d)	62.30 62.68	4.40 4.51	20.85 20.89
24 $p-Me_2N.C_6H_4$ - $C_{24}H_{24}O_3N_6$ 281 (d) 64 91 5 09 64.85 5.44 25 C_6H_5 -CH = CH- $C_{24}H_{21}O_3N_5$ 260 - 1 (d) 67.51 4.50 67.44 4.95 26 C_H-0-2- C_{24}H_2O_3N_5 285 (d) 61 60 4 37	23	<i>p</i> -MeO.C ₆ H ₄ *	$C_{23}H_{21}O_4N_5$	268 - 9 (d)	64.16 64.03	4.50 4.91	16.63 16.23
25 C_6H_5 -CH = CH- $C_{24}H_{21}O_3N_5$ 260 - 1 (d) 67.51 4.50 67.44 4.95 26 C.H.0-2- C.HQ.N_2 285 (d) 61.60 4.37	24	<i>p</i> -Me ₂ N.C ₆ H ₄ -	$C_{24}H_{24}O_{3}N_{6}$	281 (d)	64 91 64.85	5 09 5.44	19.26 18.91
26 C_{H} $\frac{1}{2}$ $\frac{1}{2$	25	C_6H_5 -CH = CH-	$C_{24}H_{21}O_3N_5$	260-1 (d)	67.51 67.44	4.50 4.95	16 50 16.38
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	26	C ₄ H ₃ 0-2-	$C_{20}H_{17}O_4N_5$	285 (d)	61.60 61.38	4.37 4.38	18.37 17.89

TABLE 2

Aldehyde Derivatives of p-Isonicotinoylaminobenzoylglycine Hydrazide

Note: d means decomposition

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Ethyl p-isonicotinoylaminobenzoyl-DL-methionate (Compound No. 27)

To a stirred solution of *p*-isonicotinoylaminobenzoic acid azide (6.5 g.) in DMF (80 ml.) was added ethyl DL-methionate (6.5 g.) in DMF (10 ml.) over a period of 30 min. at room temperature. The filtered solution after 10 hr. of stirring, was concentrated and diluted with benzene-petrol mixture (1:1) to give the required product which was filtered, washed, dried and crystalised from ethanol. m.p. 182° C. Yield: 7.8 g.; $79.8\%_{0}^{\circ}$. (Found: C, 60.32; H, 6.23; N, 10.47. C_{20} H₂₃O₄N₃S requires: C, 59.85; H, 5.74; N, 10.47%).

p-Isonicotinoylaminobenzoyl-DL-methionine (Compound No. 28)

Ethyl *p*-isonicotinoylaminobenzoyl-DL-methionate (0.5g) was hydrolysed with potassium hydroxide solution (1N, 10 ml.) and the acid isolated in the usual way. m.p. 144-5°C. Yield: 0.24 g.; 50%. (Found: N, 10.96. $C_{18}H_{19}O_4N_3S$ requires: N, 11.26%).

p-Isonicotinoylaminobenzoyl-DL-methionine hydrazide (Compound No. 29)

This was prepared by treating ethyl *p*-isonicotinoylamino-DL-methionate (8 g.) in ethanol (300 ml., 90%) with hydrazine hydrate (6 g., 99%) in the usual way. m.p. 222-23°C. Yield: 68 g.; 87.7%. (Found: C, 55.43; H, 5.65; N, 18.27. $C_{18} H_{21} O_3 N_5 S$ requires: C, 55.82; H, 5.43; N, 18.09%).

Aldehyde derivatives of p-isonicotinoylaminobenzoyl-DL-methionine hydrazide

The hydrazones were prepared in the usual manner. Table 3 contains the compounds prepared (Nos. 30 to 39) in this series.

(B) p-NICOTINOYLAMINOBENZOIC ACID, ITS GLYCINE AND DL-METHIONINE CONJUGATES AND THEIR DERIVATIVES

Ethyl p-nicotinovlaminobenzoate¹¹ (Compound No. 40)

To a stirred solution of nicotinoyl chloride in dry benzene (300 ml.), prepared from potassium nicotinate (15 g.) and thionyl cloride (30 ml.) in the usual way, was added a solution of ethyl *p*-aminobenzoate (15.4 g.) dissolved in benzene (200 ml.) during a period of 1 hr. at room temperature and the product was isolated in the usual manner. m.p. $130-1^{\circ}$ C. (Reported¹¹ m.p. 128° C.). Yield 22 g.; 89.4%. (Found: C, 66.27; H, 5.65; N, 10.07. C₁₅ H₁₄O₃ N₂ requires: C, 66 66; H, 5.22; N, 10.37%).

Т	A	B	L	Е	3
			-		

Aldehyde Derivatives of p-Isonicotinoylaminobenzoyl-DL-Methinoine Hydrazide

N CONH	CONH-CH-CONH-N=CH-R
	C112-C112.0 C 118

 $-\mathbf{R} = Aldehvde residu$

Com- R-		16-1 C		Found/Required		
No.	d Aldehyde residue	Mol. formula	m. p. °C	с	(%) H	N
30	C ₆ H ₅ -	$C_{25}H_{25}O_3N_5S$	245 (d)	63.48 63.16	4 86 5.26	14 52 14 73
31	<i>о-</i> ОН.С ₆ ^H 4 ⁻	$C_{25}H_{25}O_4N_5S$	240 (d)	61.31 61.09	5.26 5.09	13 82 14 26
32	<i>m</i> -OH C ₆ H ₄ -	$C_{25}H_{25}O_4N_5S$	263 (d)	60.73 61.09	5.35 5.09	13 96 14.26
33	<i>p</i> -OH.C ₆ H ₄ -	$C_{25}H_{25}O_4N_5S$	269 - 70 (d)	60 61 61.09	5.00 5 09	13 96 14 26
34	C5H4N-3-	$C_{24}H_{24}O_{3}N_{6}S$	237 (d)	60 36 60 50	4.70 5 04	17 40 17 65
35	C_5H_4N-4-	$C_{24}H_{24}O_3N_6S$	232 (d)	60 12 60.50	4 70 5.04	17.39 17.65
36	<i>p</i> -MeO.C ₆ H ₄ -	$C_{26}H_{27}O_4N_5S$	257-8 (d)	61 42 61.79	5.64 5.35	14.26 13.87
37	<i>p</i> -Me ₂ N.C ₆ H ₄ -37	$C_{27}H_{30}O_3N_6S$	256 (d)	62.73 62.55	5 59 5.79	16 62 16 22
38	C ₆ H ₅ -CH = CH	$C_{27}H_{27}O_3N_5S$	235-6 (d)	64.65 64.67	5.06 5.39	14 00 13.97
39	C ₄ H ₃ O-2-	$C_{23}H_{23}O_4N_3S$	206	59.27 59.34	5.11 4.95	14.70 15.05

Note: 1 means decomposition

p-Nicotinoylaminobenzoic acid^{12, 13, 14} (Compound No. 41)

This was obtained by the hydrolysis of compound No. 40 (1.g.) with potassium hydroxide solution (1N, 20 ml.). m.p. $315^{\circ}C.(d)$. (Reported m.p. $299^{\circ}C.^{12}$; 293-4°C.¹³; and $300-2^{\circ}C.^{14}$). Yield: 0.35 g.; 39%. (Found; C, 64.00; H, 4.18; N, 11.61. $C_{13}H_{10}O_3N_2$ requires: C, 64.46; H, 4.16; N, 11.56%).

p-Nicotinoylaminobenzoic acid hydrazide (Compound No. 42)

This was obtained by treating ethyl p-nicotinoylaminobenzoate (20 g.) with hydrazine hydrate (22 g.; 99%) in the usual way, m.p. 242-4°C. Yield : 13 g. ; 69%. (Found : C, 60.53 ; H, 4.45 ; N, 21.72. C, H, O, N. requires: C, 60.93; H, 4.72; N, 21.87%).

Aldehyde derivatives of p-nicotinovlaminobenzoic acid hydrazide

These were prepared in the usual manner. Table 4 contains the compounds prepared (Nos. 43 to 52) in this series.

		#	Transfer		-R = Aldeh	yde residu
Con pou No	m- R nd Aldehyde . residue	Mol. formula	m. p. °C	Fou C	nd/ Re quired (%) H	i N
43	C ₆ H ₄ -	$C_{20}H_{16}O_2N_4$	269	69 52 69 76	4 57 4 68	15.86 16 27
44	o-OH.C ₆ H ₄ -	$C_{20}H_{16}O_3N_4$	251	66 34 66 66	4 26 4 48	15. 49 15.55
45	<i>m</i> -OH.C ₆ H ₄ -	$C_{20}H_{16}O_3N_4$	304 (d)	67 18 66 66	3.99 4 48	15.46 15.55
46	<i>p</i> -OH.C ₆ H ₄ -	$C_{20}H_{1\circ}O_{3}N_{4}$	315 - 7 (d)	66.29 66.66	4.12 4 48	15.10 15 55
47	C₅H₄N-3-	$C_{19}H_{15}O_2N_5$	270	66.30 66 08	4.48 4.38	20.18 20.28
48	C5H4N-4-	$C_{19}H_{15}O_2N_5$	288	66 04 66.08	4 08 4.38	20.10 20.28
49	p-MeO.C ₆ H ₄ -	$C_{21}H_{18}O_3N_4$	285	67.57 67.37	4.88 4.85	15.27 14.96
50	$p-Me_2N.C_6H_4$	$-C_{22}H_{21}O_2N_5$	279	67.75 68 20	5.26 5.46	$18.32 \\ 18.08$
51	$C_6H_5-CH=CI$	H- $C_{22}H_{18}O_2N_4$	281	70.98 71.34	4.58 4.90	15.28 15.13
52	C ₄ H ₃ O-2-	$C_{18}H_{14}O_3N_4$	276 - 7	64.51 64.67	4.12 4.22	16.92 16.76

TABLE 4

Aldehyde derivatives of p-Nicotinoylaminobenzoic Acid Hydrazide

CONH-N=CH-R

Note: d means decomposition

Ethyl p-nicotinoylaminobenzoylglycinate (Compound No. 53)

This was obtained by condensing ethyl glycinate (5.67 g.) with p-nicotinoylaminobenzoic acid azide (9.8 g.) in DMF at room temperature in the usual way as described earlier (cf., Compound No. 14). m.p. 218°C. Yield: 9.8 g.; $82.5^{\circ}_{0.7}$. (Found: C, 62.41; H, 4.90; N, 13.29. $C_{17}H_{17}O_4N_3$ requires: C, 62.38; H, 5.23; N, 12.84 γ_0).

p-Nicotinovlaminobenzovlglycine (Compound No. 54)

This was prepered by the hydrolysis of the compound No. 53 (1 g.) with potassium hydroxide solution (IN, 20 ml) in the usual way. m.p. 295°C. (d). Yield: 0.3 g.; $32^{\circ}_{.0}$ (Found: C, 59.99; H, 4.29; N, 13.68. $C_{15}H_{13}O_4N_3$ requires: C, 60.20; H, 4.38; N, 14.04°.

p-Nirotinoylaminobenzoylglycine hydrazide (Comound No. 55)

This was prepared by refluxing ethyl p-nicotinoylaminobenzoylglyc'nate (10 g.) in ethanol (500 ml., 90%) with hydrazine hydrate (9.2 g., 59%) m.p. above 300°C. (d). Yield: 7 g.; 73.1%. (Found. C, 57.18; H, 4.52; N, 22.44. $C_{15}H_{15}O_3N_5$ requires: C, 57.50; H, 4.83; N, 22.35%).

Aldehyde derivatives of p-nicotinoylaminobenzoylglycine hydrazide

These were prepared in the usual way. Table 5 contains the compounds prepared (Nos. 56 to 65) ip this series.

Ethyl p-nicotinoylaminobenzoyl-DL-methionate (Compound No. 66)

This was prepared in the usual way by treating ethyl DL-methionate (6.98 g.) with *p*-nicotnoylaminobenzoic acid azide (7 g.) in DMF at room temperature. m,p. 187-8°C Yield: 8 g.; 76.1%. (Found: C, 60.10; H, 5.55; N, 10.72. C_{20} H₂₃ O₄ N₃ S requires: C, 59.85; H, 5.74; N, 10.4%.)

p-Nicotinoylaminobenzoyl-DL-methionine (Compound No. 67)

This was prepared by the hydrolysis of compound No. 66 (0 5 g.) with potassium hydroxide solution (1N, 10 ml.) in the usual way. m.p. 232-4°C. Yield: 0.25 g.; 51%. (Found: C, 58.41; H, 554; N, 10.99. $C_{18}H_{19}O_4N_3S$ requires: C, 57.92; H, 5.09; N. 11.26%).

p-Nicotinoylaminobenzoyl-DL-methionine hydrazide (Compound No. 68)

This was prepared in the usual way by refluxing ethyl *p*-nicotinoylaminobenzoyl-DL-methionate (6 g.) and hydrazine hydrate (4.5 g., 99%) in ethanol. m.p. softens at 175°C. and melts at 200-1°C. Yield: 4.8 g.; 90.5%. (Found: C, 55.52; H, 5.66; N, 18.05. $C_{18}H_{21}$, O_3N_5S requires: C, 55.82; H, 5.43; N, 18.09%).

	Aldehyde derivatives of p-Nicotinoylaminobenzoylglycine Hydrazide							
			·CONH-CH2-	CONH-N	'=CH−R -R=Aldehv	de residue		
Co pou N	m- R- Ind Aldehyde o. residue	Mol. formula	m. p. °C	Fou	nd/Required (%) H	N		
56	C₅H₅-	C ₂₂ H ₁₉ O ₃ N ₅	260	65.54 66.83	4.74 4.77	16.94 17.45		
57	o-OH.C ₆ H ₄-	$C_{22}H_{19}O_{4}N_{5}$	261	62.90 63.30	4.63 4 59	16.44 16.78		
58	<i>m</i> -OH.C ₆ H ₄ -	C ₂₂ H ₁₉ O ₄ N ₅	310 (d)	62.90 63.30	4.49 4.59	16.44 16.78		
59	<i>p</i> -OH.C ₆ H₄-	$C_{22}H_{19}O_4N_5$	310 (d)	63.09 63.30	4.96 4.59	16.44 16.78		
60	C ₅ H ₄ N-3-	$\mathrm{C}_{21}\mathrm{H}_{18}\mathrm{O}_{3}\mathrm{N}_{6}$	275 (d)	62.22 62.68	4.80 4.51	21.24 20.89		
51	C ₅ H ₄ N-4-	$C_{21}H_{18}O_{3}N_{6}$	254 (d)	62.32 62.68	4.74 4.51	21.08 20.89		
52	p-MeO.C ₆ H ₄ -	$C_{23}H_{21}O_{4}N_{5}$	266-7(d)	64.39 64.03	4.49 4.91	16.07 16.23		
13	<i>p</i> -Me ₂ N.C ₆ H ₄ -	C ₂₄ H ₂₄ O ₃ N ₆	266-(d)	64.55 64 85	5.50 5.44	19 21 18.91		
54	$C_6H_5-CH = CH-$	$C_{24}H_{21}O_{3}N_{5}$	263-4	67.38 67.44	5.16 4.95	16.08 16.38		
5	C ₄ H ₃ O-2-	$C_{20}H_{17}O_{4}N_{5}$	282 (d)	61·10 61.38	4.58 4.38	17.45 17.89		

TABLE	5
-------	---

Note: d means decomposition

ldehyde derivatives of p-nicotinoylaminobenzoyl-DL-methionine hydrazide

The hydrazones were prepared in the usual manner. Table 6 contains e compounds prepared (Nos. 69 to 78) in this series.

TABLE 6

Aldehyde derivatives of p-Nicotinoylaminobenzoyl-DL-Methionine Hydrazide

delivatives of printerinely mini-	
	-CH-CONH-N=CH-R
	CH2-CH2-S-CH3

-	11		-C	H.	
~	r .	2-0	\sim		

- R = Aldehyde residue

Com-	R			Four	d/Requited		
poune No.	d Aldehyde residue	Mol, formula	m.p. °C	С	(%) <u>H</u>	N	
69	C ₆ H ₅ -	C ₂₅ H ₂₅ O ₃ N ₅ S	265 (d)	62.88 63.16	5.24 5 26	14.33 14 73	
70	<i>o</i> -OH.C ₆ H ₄ -	C ₂₅ H ₂₅ O ₄ N ₅ S	5 238 (d)	60 68 61.09	4.79 5.09	14 27 14.26	
71	<i>m</i> -OH,C ₆ H ₄ -	C ₂₅ H ₂₅ O ₄ N ₅ S	S 269 - 70 (d)	61 08 61.09	4 75 5.09	14.07 14.26	
72	<i>p</i> -OH.C ₆ H ₄ -	$C_{25}H_{25}O_4N_5S$	262-3 (d)	60 91 61.09	5.26 5.09	13.86 14 26	
73	C ₅ H ₄ N-3-	C ₂₄ H ₂₄ O ₃ N ₆ S	228-9	60.16 60 50	5 46 5 04	17.78 17.65	
74	C ₅ H ₄ N-4-	$C_{24}H_{24}O_{3}N_{6}S$	233	60.37 60 50	5.32 5.04	17.78 17 65	
75	<i>p</i> -MeO.C ₆ H ₄ -	$C_{26}H_{27}O_4N_5S$	228-9 (d)	61.80 61.79	5.37 5.35	13.68 13.87	
76	p-Me ₂ N.C ₆ H ₄ -	C ₂₇ H ₃₀ O ₃ N ₆ S	5 243-5 (d)	62.92 62 55	5 97 5.79	16.34 16 22	
77	$C_{6}H_{5}$ -CH = CH-	C ₂₇ H ₂₇ O ₃ N ₅ S	5 238-9 (d)	64.52 64.67	5 27 5.39	14.26 13.97	
78	C4H3O-2-	C ₂₃ H ₂₃ O ₄ N ₅ S	5 188	59.58 59.34	5 01 4.95	14.80 15.05	

Note: d means decomposition

(C) *p*-PICOLINOYLAMINOBENZOIC ACID, ITS GLYCINE AND DL-METHIONINE CONJUGATES AND THEIR DERIVATIVES

Ethyl p-picolynoylaminobenzoate (Compound No. 79)

To a suspension of potassium picolinate (10 g.) in dry benzene (60 ml.). cooled to 0°C., was added slowly redistilled thionyl chloride (20 ml.) over a period of 10 min. After the addition, the reaction mixture was kept as such for half an hour and then the benzene and excess of thionyl chloride were removed by distillation under suction.

To a stirred solution of this crude picolinoyl chloride in dry benzene (150 ml.), was added a solution of ethyl *p*-aminobenzoate (10.25 g.) in dry benzene (150 ml.), and the product isolated in the usual way. m.p. 135°C. Yield: 13.5 g.; 82.5%. (Found: C, 67.01; H, 479; N, 10.32. C, 51 H₄O₃ N₂ requires: C, 66.66; H, 5.22; N, 10.36%).

p-Picolinoylaminobenzoic acid (Componnd No. 80)

This was prepared by the hydrolysis of the compound No. 79 (1 g.) with potassium hydroxide solution (1N, 20 ml.) in the usual way. m.p. 265-7°C. Yield: 0.12 g.; 14%. (Found: N, 11.61. $C_{13} H_{10} O_3 N_2$ requires: N, 11 56%.)

p-Picolinoylaminobenzoic acid hydraziae (Compound No. 81)

This was obtained by refluxing ethyl *p*-picolinoylaminobenzoate (20 g) and hydrazine hydrate (22 g., 99%) in ethanol in the usual way. m.p. 247-8°C. Yield: 15 g.; 79.1%. (Found: C, 60.61; H, 4.72; N, 22.18. C_{13} H₁₂ O₂ N₄ requires C, 60.93; H, 4.72; N, 21.87%).

Aldehyde derivatives of p-picolinoylaminobenzoic acid hydrazide

The hydrazones were prepared in the usual way. Table 7 contains the compounds prepared (Nos. 82 to 87) in this series.

Ethyl p-picolinoylaminobenzoylglycinate (Compound No. 88)

This was prepared by condensing ethyl glycinate (8.1 g.) with p-picolinoylaminobenzoic acid azide (14 g.) in DMF at room temperature in the usual way. m.p. 172-3°C. Yield: 15 g.; 87.5%. (Found: C, 62.05; H, 5.17; N, 12.53. C_{17} H₁₇ O₄ N₃ requires: C, 62.38; H, 5.23; N, 12.84\%).

	Aldenyde	-CONH-N	-CH-R	riyorazide	
				-R=Aldeh	yde residue
Con poun No.	d Aldehyde residue	Mol. formula m. p. °C	For	und/Required (°/0) H	N
82	C ₆ H ₅ -	C ₂₀ H ₁₆ O ₂ N ₄ 247-8	70.11 69.76	4.60 4.68	15 86 16.27
83	o-OH.C ₆ H ₄ -	$C_{20}H_{16}O_{3}N_{4}$ 245-6	66.53 66.66	4.25 4.48	15.23 15.55
84	m-OH.C ₆ H ₄ -	$C_{20}H_{16}O_{3}N_{4}$ 270-1 (d)	66.22 66.66	4.45 4.48	15.17 15 55
85	<i>p</i> -OH.C ₆ H ₄ -	$C_{20}H_{16}O_{3}N_{4}$ 311-2 (d)	66.65 66.66	4.41 4.48	15.19 15.55
86	C ₅ H ₄ N-3-	$C_{19}H_{15}O_{2}N_{5}262-4$	66.03 66.08	4.13 4.38	19 97 20.28
87	C ₅ H ₄ N-4-	$C_{19}H_{15}O_{2}N_{5}269-70$	66.51 66.08	3 95 4.38	19.92 20.28

TABLE 7

Aldehyde derivatives of p-Picolinoylaminobenzoic Acid Hydrazide

Note: d means decomposition

p-Picolinoylaminobenzoylglycine (Compound No. 89)

This was prepared by the hydrolysis of the compound No. 88 (0.5 g.) with potassium hydroxide solution (1N, 10 ml.) in the usual way. m.p. 235°C. (d). Yield: 0.24 g.; 50%. (Found: C, 60.13; H, 4.63; N, 14.21. $C_{15} H_{13} O_4 N_3$ requires: C, 60.20; H, 4.38; N, 14.04%).

p-Picolinoylaminobenzoylglycine hydrazide (Compound No. 90)

This was prepared from ethyl *p*-picolinoylaminobenzoylglycinate (12 g.) and hydrazine hydrate (11 g., 99%) in the usual manner. m.p. 257°C. Yield: 10.3 g.; 91%. (Found: C, 57.97; H, 4.79; N, 22.15. $C_{15}H_{15}O_3N_5$ requires: C, 57.50; H, 4.83; N, 22.35%).

Aldehyde derivatives of p-picolinoylaminobenzoylglycine hydrazide

The hydrazones were prepared in the usual way. Table 8 contains the compounds prepared (Nos. 91 to 96) in this series.

					=Aldehy	de resid
Com poun No,	d Aldehyde residue	Mol. formula	m,p,°C	Fou C	nd/Required (%) H	I N
91	C ₆ H ₅ -	$C_{22}H_{19}O_{3}N_{5}$	265	65.70 65.83	4 42 4 77	17 9 17.4
92	o-OH.C ₆ H ₄ -	$C_{22}H_{19}O_4N_5$	263-4(d)	63.10 63 30	4 69 4.59	17.2 16.7
93	m-OH.C ₆ H ₄ -	$C_{22}H_{19}O_{4}N_{5}$	255	6 3 11 63.30	4.19 4 59	16.92 16.73
94	p-OH.C ₆ H ₄	$C_{22}H_{19}O_4N_5$	270-1(d)	62.81 63.30	4 58 4.59	16.81 16.78
5	C ₅ H ₄ N-3-	$C_{21}H_{18}O_{3}N_{6}$	278 (d)	63.19 62.68	4 78 4 51	20-74 20.89
6	C ₅ H ₄ N-4-	$C_{21}H_{18}O_{3}N_{6}$	255 (d)	62.40 62.68	4 37 4 51	20.71 20 8 9

TABLE 8

Aldehyde derivatives of p-Picolinoylaminobenzoylglycine Hydrazide

Note: d means decomposition

Ethyl p-picolinoylaminobenzovl-DL-methionate (Compound No. 97)

This was prepared by treating *p*-picolinoylaminobenzoic acid azide (6 g.) with ethyl DL-methionate (6 g.) in DMF at room temperature in the usual way. m.p.:148-49°C. Yield: 6.5 g.: 72.2%, (Found: C, 59.53; H, 583; N, 10.25. C_{20} H₂₃ O₄ N₃ S requires: C, 59.85; H, 5.74; N, 10.47%).

p-Picolinoylaminobenzoyl-DL-methionine (Compound No. 98)

Hydrolysis of the compound No. 97 (0.5 g.) with potassium hydroxide solution (1N, 10 ml.) in the usual way gave the required acid. m.p. 192°C. Yield: 0.16 g.; 33%. (Found: N, 11.32. $C_{18}H_{19}O_4N_3S$ requires: N, 11.26%).

p-Picolinoylaminobenzoyl-DL-methionine hydrazide (Compound No. 99)

This was obtained in the usual way by refluxing ethyl *p*-picolinoylaminobenzoyl-DL-methionate (12 g.) with hydrazine hydrate (9 g., 99%) in ethanol. m.p. 206-7°C. Yield: 10.1 g.; 87°_0} . (Found: C, 55.69: H, 5.27; N, 17.72. C_{18} H₂₁ O₃ N₅ S requires: C, 55.82; H, 5.43; N, 18.09%).

Aldehyde derivatives of p-picolinoylaminobenzoyl-DL-methinonine hydrazide

CONH

The hydrazones were prepared in the usual way. Table 9 contains the compounds prepared (Nos. 100 to 105) in this series.

TAB	LE	9	

Aldehyde derivatives of p-Picolinoylaminobenzoyl-DL-Methionine Hydrazide

CONH-CH-CONH-N=CH-R

ĊH₂.CH₂.S.CH₃		
	$-\mathbf{R} = \mathbf{A} \mathbf{I} \mathbf{d} \mathbf{e} \mathbf{h} \mathbf{v} \mathbf{d} \mathbf{e}$	residue

Com				Fou	nd/Required	
No.	d Aldenyde residue	Moi. formula	m. pC	с	(*/~) H	N
100	C ₅ H ₆ -	C ₂₅ H ₂₅ O ₃ N ₅ S	210	63.59 63 16	5.47 5.26	14 67 14.73
101	0-OH.C ₆ H₄-	C ₂₅ H ₂₅ O ₄ N ₅ S	215-6	60.59 61.09	4.95 5.09	14.70 14 26
102	m-OH.C ₆ H ₄ -	$C_{25}H_{25}O_{4}N_{5}S$	210	60.67 61.09	5 22 5 09	14.56 14.26
103	<i>p</i> -OH.C ₆ H ₄ -	$C_{25}H_{25}O_{4}N_{5}S$	248	61 31 61.09	5.14 5.09	14.32 14.26
104	C ₅ H ₄ N-3-	$C_{24}H_{24}O_{3}N_{6}S$	228 - 9	60.46 60.50	4.62 5.04	17.68 17.65
105	C ₅ H ₄ N-4-	$C_{24}H_{24}O_{3}N_{6}S$	241 – 2 (d)	60.19 60.50	5.42 5.04	17.26 17.65

Note: d means decomposition

Results of in vitio Screening

A total of 105 compounds have been synthesised and tested for *in vitro* antituberculosis activity by the surface culture technique, using the virulent strain of *Mycrobacterium tuberculosis* $H_{37}R_{o}$. Out of this, 19 compounds have shown complete inhibition of growth of the organism at a dilution of I in 10,000 (100 γ /ml.). Twenty compounds were partially inhibitory at this dilution. Among the 19 active compounds, some have inhibited the growth even at higher dilutions. The screening data of all the compounds are summarised in Table 10.

The following generalisations are made from the results of antituberculosis tests.

- (i) p-Pyridoylaminobenzoic acids and their amino acid conjugates are inactive in general.
- (ii) The ethyl esters of all the three p-pyridoylaminobenzoic acids are highly active. Conjugation with either glycine or DLmethionine destroys the activity.
- (iii) All the three hydrazides of p-pyridoylaminobenzoic acids are active. While the hydrazides of the glycine conjugates are active, those of DL-methionine are inactive.
- (iv) Formation of hydrazones destroys the activity in most of the cases.
- (v) Esters, acids, hydrazides and hydrazones of picolinoyl series are more active than the corresponding derivatives from nicotinoyl and isonicotinoyl series.

In Vivo Activity

Among the compounds tested, ethyl *p*-isonicotinoylaminobenzoate (Compound No. 1), ethyl *p*-nicotinoylaminobenzoate (Compound No. 40) and ethyl *p*-picolinoylaminobenzoate (Compound No. 79) were found to exhibit *in vitro* antituberculosis activity equivelent to those of *p*-aminosalicylic acid, streptomycin and isoniazid. They were therefore tested for *in vivo* potency in experimental tuberculosis of mice by the drug-diet method and compared with isoniazid treated animals. None of the compounds was found to be active in vivo. They neither showed any survivors nor prolonged the life span of the treated mice compared to the controls.

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Results of Screening for	Inhibitio u	n of Gro	wth in Pi	IN OF My	cobacteri	ım Tuber	culosis H _i	$r R_2$		
punod	voznadonimskyonitozicost-q seid derivatives	p-lsonicotinoylaminobenzoy glycine deruatives	p.1soniosinekeninosines DL-methionne detivesuves	pNicotinoylaminodenzoic acid derivatives	P.N icotinoylaminobenzoyl- glycine derivatives	P.Vişotinoylamınobenzoyl- D.L-methionine derivatives	p-Pisolinaylaminobenzoic acid derivatives	-Picolinoylaminobenzoyl- glycine derivatives	-Picolinnsylaminobenzoyl- DL-methionine derivatives	Total number of active compounds
	đ	+	+	IJ	+	+	10	1	 +	4
	+	+	+	÷	+	+	+!	+	+	ļ
	ŧ	+1	÷	ł	ŧ	÷	+4	1	+	S
HTIW SEV										
	+	-+1	÷	÷	+	+	+1	44	+	0
ldehyde	+	+1	+	+	1	÷	+H	· +!	+	
ıldehyde	+	+1	+	÷	+	÷	+	-11	+	0
ldehyde		+1	÷	÷	-}-	+	+1	1	+	Г
	+	+1	~ -	÷	÷	÷	+)	ł	+	
/de	÷	+1	+	I	+	+	I	I	+	ŝ
ldehyde	+	+1	+	+	÷	+	×	×	×	0

TABLE 10

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 <i>p</i>-Dimethylamino bei Cinnamaldehyde 2-Futanaldehyde 	nzaldehyde	+ + +	++ + + ++	+ + +	+ + +	+	+ +# +	x × ×	x	x x x	
Compounds in each Tota Serial	series : I No. Nos.	13 1-13	13 14-26	13 27-39	13 40-52	13 53-65	13 67-78	9 79-87	9 88-96	97-105	
Active compounds in Tota Serial	each serie 11 No. Nos.	1, 3	00	0 0	3 40, 42, 48	4 55, 57 63, 65	1 77	4 79, 80, 81, 87	5 88, 90, 94, 95, 96	0 0	→ 6[
Legend	dicates comm		1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 -	-							
= inc	dicates comp	lete in	hibition	at I:1 at I:1	u,uuu ai million	lution dilutio	ť				
≡ inc	dicates comp	lete in	hibition	at 1:1	0 millio	n dilutio	л.				
+ + ∓	dicates partia	al inhit	ition at	1:100	,000 dih	ttion.					
± ii	dicates partie	l inhił	vition at	1:10,0	000 dilut	ion.		-			
+ inc	dicates no in	hibitio	n at 1:	10,000	dilution.						
× coj	mpounds not	t prepa	red.								
Ethyl-p-an	ninobenzoate	inhibi	ts the g	rowth c	of <i>M.t.</i> i	n vitro a	1:10	million	dilution		
p-Aminob inactive at this	enzoic acid a e at 1 : 10,000 dilution.	and all dilutic	the ald m. <i>p</i> -I	ehydes, Jimethy	exceptir Iaminob	ıg <i>p</i> -din enzaldel	iethylan iyde she	ninobenz ows part	zaldehyd ial inhib	e are ition	
Note : Out of 105 compour	nds (<i>i.e.</i> , 9 este ibited the grov	rs,9asci wth:ofA	ds, 9 hyd. 1.r. at 1:	razides a 10,000 di	nd 78 hydi hution. T	fazones) s wentv co	ynthesise mnounde	d and ic were po	sted, 19 cc	4 spunoduc	ave

dilution. Among the active compounds some have shown inhibition even at higher dilutions.

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