Senna-its chemistry, distribution and pharmaceutical value *

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Abstract

The active principles in senna are anthraquinone derivatives, namely, rhein, aloe-emodin, chrysophanic acid and physcion present both in free and in the form of glycosides. Sennosides A and B present in senna are the dianthrone derivatives of rhein with two glucose units. The leaves contain maximum sennosides at the time of flowering and the hand-picked pods are superior to black pods with regard to sennosides content. The anthraquinone derivatives are biosynthesized via (i) The acetate-polymalonate pathway and (ii) from shikimic acid. The cathartic action of senna is attributed to the presence of anthraquinones and/or related compounds. Preparations involving leaves and pods of *Cassia* species are described.

Key words: Anthracene compounds, sennosides, rhein-dianthrone glycosides, cathartic action, medicaments.

1. Introduction

Senna is an important member of the vegetable drugs containing anthracene group of derivatives. Some of the other members of this group are Aloes, Rhubarb, Cascara sagrada and Frangula. All these are useful as purgatives and the purgative effect is due to their irritant action on the bowel¹. Records of 9th century A.D. show that senna was employed by the great Arab physicians². They introduced the drug into Western Europe. Senna purges gently and is safe for all ages. Senna specifically denotes drug preparations from dried leaflets of Cassia acutifolia Delile (Alexandrian Senna) and Cassia angustifolia Vahl (Tirunelveli Senna). However, other species of Cassia are also economically valuable. Some Cassia species have been reported to possess antibiotic or curariform activity³. It has also been reported that cassias are used to treat constipation, diabetes and haemoglobin disorders³.

C. acutifolia grows naturally and is also cultivated in the middle and upper Nile countries. C. angustifolia is commonly cultivated in South India and coastal arid region of Kutch⁴. These two species are also successfully domesticated in other parts of the world; C. angustifolia and C. acutifolia in Imperial Valley of California⁵,

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C. angustifolia in Northern Brazil⁶ and C. acutifolia in Soviet Union⁷. From India six million rupees worth of senna drugs, pods and leaves are exported annually⁴. Senna leaf is sold in European market as Senna Tea. Naturally enough, the sophisticated customers there prefer to buy only healthy and bright coloured leaves.

2. Anthracene derivatives isolated from senna

Table I gives the information regarding the number of anthracene compounds isolated and identified from leaves and pods of senna. The active principles in Senna are

Table I

Cassia acutifolia Delile		Cassia angustifolia Vahl		
(Alexandrian Senna)		(Tirunelveli Senna)		
Leaves	Pods	Leaves	Pods	
1. Sennosides	Sennosides	Same as in	Sennosides A and B $1 \cdot 2$ to $2 \cdot 5\%$	
A and B	A and B	Alexandrian		
2 to 3%	2.5 to 4.5%	Senna leaves		
 A second s	A and B	Alexandrian		

Anthracene derivatives isolated in Senna

- 2. Sennosides C and D
- Rhein anthrone 8-glycoside
- Rhein
 8-glycoside
- Rhein-8-diglycoside
- Aloe-emodin-8-glycoside
- 7. Aloe-emodin anthronediglycoside
- 8. Free rhein and aloe-emodin
- 9. 30-40% water soluble extractive

Sennosides C and D

Rhein anthrone-8-glycoside

Rhein 8-glycoside

Rhein-8-diglycoside

Aloe-emodin-8-glycoside

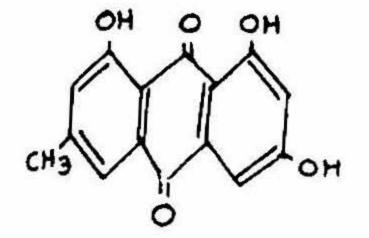
Aloe-emodin anthrone diglycoside

26-31% water soluble extractive Same as No. 2-1 m pods of Alexandrian Senna.

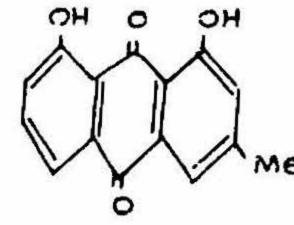
21 to 26% watersoluble extractive

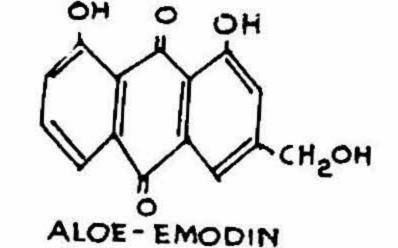
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anthracene derivatives and are present both in free and in the form of glycosides. structures of some important anthracene derivatives are given in Fig. 1. The



FRANGULA- EMODIN

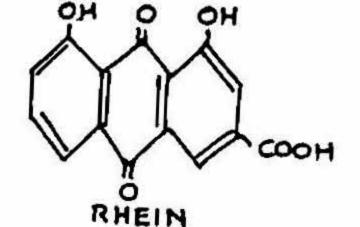




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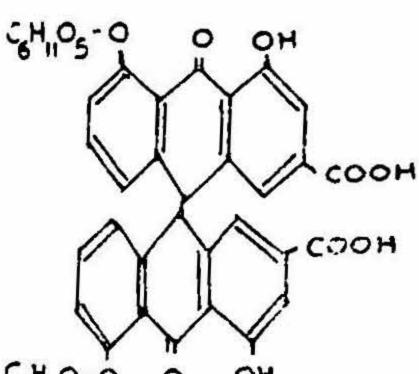
PHYSCION

OH



OH Me Me

CHRYSOPHANIC ACID



 Sennoside-A: Dextrorotatory aglycon sennidin. A and D = glucose.

0 Me

11. Sennocide -B: Intramolecularly compensated mesosennidin B and D-glucose.

1

5 H 0 0 0H

ANTHRAQUINONE GL'/COSIDE

Fig. 1. Important anthracene compounds in senna.

Senna contains rhein, aloe-emodin in the free and in the form of glycosides. The most important constituents, however, are sennosides A and B. Stoll et al.⁸ isolated these compounds in crystalline form from leaves of Senna. Sennosides A and B are dianthrone derivatives of rhein with two glucose units. The rhein dianthrone compounds are Sennidins A and B. The presence of third dianthrone compound is reported^{9,10}. This was called sennoside C and is shown to be 8,8'-diglycoside of rhein, aloe-emodin dianthrone. Apart from these compounds, presence of two more dianthrone derivatives has been established. Lemli and Cuveele¹¹ isolated diglycoside of aloe-emodin dianthrone from senna leaves by column chromatography on silica gel and polyamide. The aglycon is levorotatory and the glucose units are fixed at position 1 or 8 and 1' or 8'. Reidin 'A', a heterodianthrone is demonstrated to be present in senna leaves¹² (Paper Chromatography studies). Fairbairn and Shreshta¹³ confirmed the presence of a highly

water soluble glycoside based on a reduced form of aloe-emodin. A number of glycosides, both mono and polyglycosides of rhein, rhein-anthrone, aloe-emodin, have been isolated and identified in *C. acutifolia*^{14,15}. Senna leaf also contains kaempferol, isorhamnatin, myricyl alcohol, salicyclic acid, phytosterolin, mucilage, resin, chryso. phanic acid and calcium oxalate¹⁶.

3. Other Cassia species possessing anthracene derivatives

Table II gives different *Cassia* species found to contain anthracene derivatives. In all 26 species of *Cassia*, from different countries, have been studied. Most of these contained anthracene derivatives, both in free and in the form of glycosides. Only a few of them other than *C. angustifolia* and *C. acutifolia* are reported to be used as drugs for cathartic activity. Of these mention may be made of *C. sieberiana*^{46,47} from Tropical Africa possessing mild purgative activity which is also used as a diuretic. *C. podocarpa* from Senegal^{35,43}, *C. fistula* from Brazil²⁷ and *C. alata*^{17,18} leaves possess activity similar to that of senna. *C. fistula*²⁹ is grown in India and medicinal properties have been attributed to nearly all parts of the tree. Of these fruits are important. The pulp from fruits known as *Cassia* pulp is a known laxative. It is used mixed with senna leaves. *C. obovata* and *C. sofora* grown in Soviet Union have same efficiency as that of Alexandrian Senna⁷. *C. podocarpa* from French West Africa is also reported to contain anthraquinone derivatives in leaves closely related to official senna⁴³.

4. Growth studies and sennosides distribution

Studies undertaken to determine the optimum conditions for growth, to obtain a crop having maximum amount of active principles and thus ensure maximum returns have indicated the following general conclusions:

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- (i) Anthraquinone glycosides tend to accumulate mostly in September-October⁵².
- (ii) The sennosides of leaves are at maximum at 45 days after germination of seeds and then progressively decrease with the maturation of pods⁴.
- (iii) Sennosides in pods are at maximum when total seed weight per pod is 22-30%⁴.
- (iv) The leaves contain maximum sennosides at the time of flowering in C. acutifolia⁵³.

The sennosides content of different grades of leaves and pods and that of flowers and leaf rachis of *C. angustifolia* are given in Table III⁵⁴. The leaf rachis had sennosides content of 1.94%. Flowers had more sennosides than different leaf grades. Hand-picked pods (seed content 6.44%) had more sennosides than black pods (seed content 12.05%). Leaf samples analysed for sennosides from various stages of growth of senna plants, namely, at non-flowering, flowering, immature pods and mature pods indicated the following results. Leaves collected during flowering stage had maximum sennosides. The content started decreasing as soon as pods started forming and it is at minimum when the plants had mature pods.

51. No.	Name of species	Active plant part	Active principle isolated	Content	Other uses	Ref.
1.	C. alata		Rhein and other anthra- quinone	2 <u></u>		17, 18
2.	C. auriculata	Leaves	emodin			19
3.	C. bicapsu- laris (Brazilian)	Fruits and Leaves		20% of the hydroxy anthracene content of C. alata		20
4.	C. chryso- carpa (Brazilian)	Epicarp Mesocarp		65% of the hydroxy anthracene content of C. alata 20% of the hydroxy anthracene content of C. alata		20
5	6. C. corymbosc	z Leaves		0.4% anthraquinone		21
ć	5. C. fistula (Somaliland, Bangladesh, Brazilian)		Free and combined rhein sennosides A and B, senni- din like compounds	 (i) 0.117% free rhein (ii) 0.229% sennidin like compound (iii) 1.15% combined hy-droxy methyl anthraquinone compounds (iv) 1.1% rhein 		22–29
	7. C. hoffman- seggi (Brazilian)	Leaves		70% hydroxy anthra- cene content of C. alata		20

Anthracene derivatives from other Cassia species

Table II

						<u> </u>
Sl. Name of No. species	Active plant part	Active principle isolated	Content	Other uses	Ref.	TOT
8. C. jaegeri	Leaves	Heterosides of chrysophanol, emodol and physcion			30	
9. C. lentophylla		Positive Borntraeger reaction	() ()		31	
10. C. marilandica (American Senna)	2 Leaves	 (1) Chrysophanol, physcion and their glycosides (2) Heterosides of emodal of the glucofragerlin type (one a and other β) 	Trace		32	
11. C. mimosoide.	s Roots, seeds and leaves	Emodin and emodin glyco- side, physcion and emodic acid	*2 *2		33, 34	
12. C. migricans	Leaves and pods	Mainly non-rhein derivatives			35	
13. C. nodose	Flowers	Nodososide—a new anthra- quinone glycoside			36	C. D. C. DEN
14. C. obovata (Egypt, USS	Leaves and R) pods.	Rhein, aloe-emodin, sennidins, reduced aloe- emodin, chrysophanols, sennosides A and B	7 3		7	
15. C. obtusifor	lia —	Gluco-obtusifolin, gluco- aurantio-obtusin	<u></u>		38	
16. C. occiden	talis Roots and seeds	Physcion, chrysophanol, rhein, aloc-emodin	1.9% free and 4.5% total anthraquinones	Seeds used in treating skin infection like ring work	39-41	

17. C. ovata (Tashkent)	Leaves	Emodin anthrone	0.8% hydroxy methyl anthraquinone		7, 42	
18. C. podocarpa (Senegal)	Leaves	Sennosides A and B, rhein anthrone glycoside, rhein and rhein glycoside	0.2% anthraquinone derivatives		35-43	
19. C. reticulata	Flowers	Rhein, aloe-emodin, emodin, chrysophanol		Possesses antibiotic activity	44, 45	
20. C. siberiana (Africa)	Fruits and leaves	Rhein and rhein-8-glycoside	Anthraquinone deri- vatives particularly abundant in leaves	Used as diuretic and is mild purgative	46, 47	
21. C. siamea		Siameanin dimer of chrysophanol, physcion and rhein			48,85	CHEMIST
22. C. singue- anna	Roots and seed	Chrysophanol, physcion and two other anthraquinones			49	TRY OF
23. C. speciosa	Roots and seed	Gives a weekly positive, Borntraeger reaction			31	F SENNA
24. C. splendida (Brazilian)	Fruits		20% hydroxy anthra- cene content of C. alata		20	-
25. C. tora (Vietnam)	Seeds	Chrysophanol, physcion, emodin, aloe-emodin and rhein and glycosides of some of these			41, 50	
26. C. torosa		Emodin-anthrone	•		51	

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Sennosides content in C. angustifolia (Samples from Tuticorin area of Tirunelveli District of Tamil Nadu)

SI.	Category of sample	ple Sennosides % expressed as - sennosides B	Average s	With	
No.			Length (cm)	Breadth (cm)	Weight of 100 leaves, pods or seeds (g)
1.	Leaves grade I	1.68	3.5	1.3	3.0
2.	Leaves grade II	1-91	3.4	0.8	2.3
3.	Leaves grade III	2.18	2.8	0.7	 1 · 6
4.	Leaves grade IV	1.88	2.2	0.5	0.7
5.	Leaves grade V	1 • 90	1.3	0.3	0.3
6.	Ungraded	2.10			
7.	Leaf rachis	1.94			
8.	Flowers	2.60	10 <u>1 - 10</u> 101 - 1070-		_
9.	Hand-picked pods	3.68	4.4	1.6	12.3
10.	Seed from above pods	0.22			0.8
11,	Black pods	2.59	4.6	1.6	15-4

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Fairbairn and Shreshtha⁵⁵ studied the distribution of anthraquinone glycosides in senna in order to understand the pathways of anthracene derivatives formation in *Cassia*. According to their findings, senna seeds contained no anthraquinone, but shortly after germination chrysophanol, then aloe-emodin and finally rhein are formed in the young plant. In the presence of adequate light, glycosylation follows and significant quantities of glycosides appear in young leaves. They are translocated to flowers and ovaries where they accumulate. During the pod development the content of anthraquinone glycosides in the pericarp decreases gradually, with the maturation of pods, first aloe-emodin and then rhein glycoside.

The anthraquinonic constituents of 21 species of *Cassia* studied by TLC technique⁴ revealed the following: (a) Different parts of the same species always contained the same group of anthracenic aglycones; (b) In *C. occidentalis*, the young roots contained chrysophanol and emodin, while, older roots contained physicon, suggesting that physicon would be the final product, and (c) The formation of aloe emodin and rhein

has been explained as taking place as a result of successive oxidation of CH_3 group in position 3 in chrysophanol. The formation of emodin could not be satisfactorily explained, but once formed, emodin on methylation at position 3 would yield physicon.

5. Biosynthesis of anthraquinone derivatives

The formation of anthraquinones in plants other than those belonging to *Cassia* species has been studied in considerable detail. Biosynthesis of anthraquinone derivatives has been shown to take place in two different pathways.

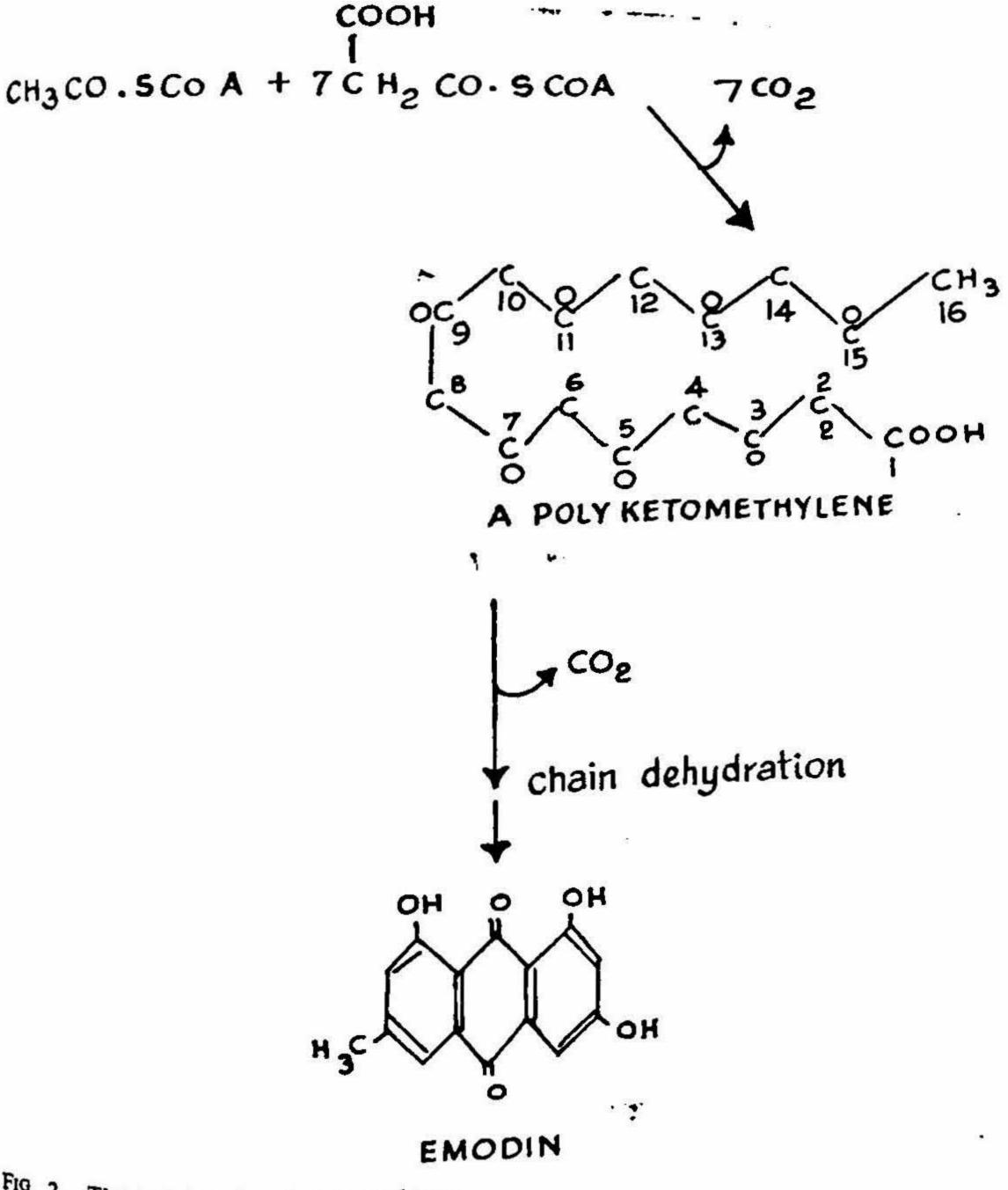


Fig. 2. The acetate-polymalonate pathway.

(i) The acetate polymalonate pathway

In this pathway, anthraquinone derivatives are believed to be formed in the following

Emodin, its dimer skyrin and islandicin are formed in this manner in Penicillium islandicum⁵⁷. The stages where the water molecules and CO₂ are lost from the poly. ketomethylene chain are not known. This type of process also goes on in plants. Leistner and Zenk⁵⁸ fed young growing leaves of Rumex alpinus through the cut ends with $(\pm)1$, 2-14C shikimic acid, $(\pm)2-14C$ mevalonate, $(\pm)1-14C$ acetate and 2-14C acetate in complete darkness. After 24 hours, chrysophanol isolated from leaves, purified and subjected to degradative studies, using alkaline H₂O₂. No incorporation of labelled shikimic acid and labelled mevalonate into the emodin type anthraquinones occurred. 1-14C and 2-14C acetates are, however, incorporated into chrysophanol. The distribution of radioactivity, in the oxidation and degradation products of labelled chrysophanol isolated, is in complete agreement with the acetate-polymalonate pathway. The alternate labelling of the chrysophanol molecule⁵⁹ after 1-14C or 2-14C acetate feeding, strongly supports the view that acetate units are linked together by way of acetyl Co A and malonyl Co A. Cyclization of the polyketide unit most probably gives rise to an anthraquinone precursor which in several steps is transformed to chrysophanol. Similar conclusions were arrived at in the studies conducted in Rumex obtusifolius⁶.

Biosynthesis of anthraquinones starting from shikimic acid (ii)

Universally labelled shikimic acid (U-14C) is specifically incorporated into alizarin and purpurin in Rubia tinctorum⁶¹ (Fig. 3).

Thus the existence of a biosynthetic pathway for the formation of anthraquinones entirely different from the acetate-polymalonate route was established. The carboxyl carbon of shikimic acid was incorporated into the anthraquinone nucleus. The same thing happens when naphthoquinones are formed from shikimic acid in bacteria and higher plants. 2-carboxy-1,4-naphthoquinol was in fact formed from shikimic acid and 2-ketoglutarate in excised shoots of 6 to 8 weeks old Impartiens balsamina⁵². The ring junction of the naphthoquinone has been shown to originate from the ethylenic carbons of shikimic acid⁶³ in Mycobacterium phlei. The carbon atoms 2, 3, 4 from 2-ketoglutarate are incorporated into 2, 3, 4 positions of the naphthoquinol⁶².

Since COOH carbon of shikimic acid was incorporated into the anthraquinone nucleus and is also incorporated into the naphthoquinone nucleus⁶², it was thought that alizarin is formed alizarin is formed via naphthoquinone in Rubiaceae plants⁶⁴. 1,4-naphthoquinone of l (1,4-14C and 2, 3, 9, 10-14C) was synthesized and supplied to the root system of ly vear old Rubia tion to the root system of ly year old Rubia tinctorum plants. Radioactive alizarin was isolated and degradation studies of the same alizarin was anthrastudies of the same showed that 1,4-naphthoquinone was incorporated into the anthra-quinone nucleurs quinone nucleus.

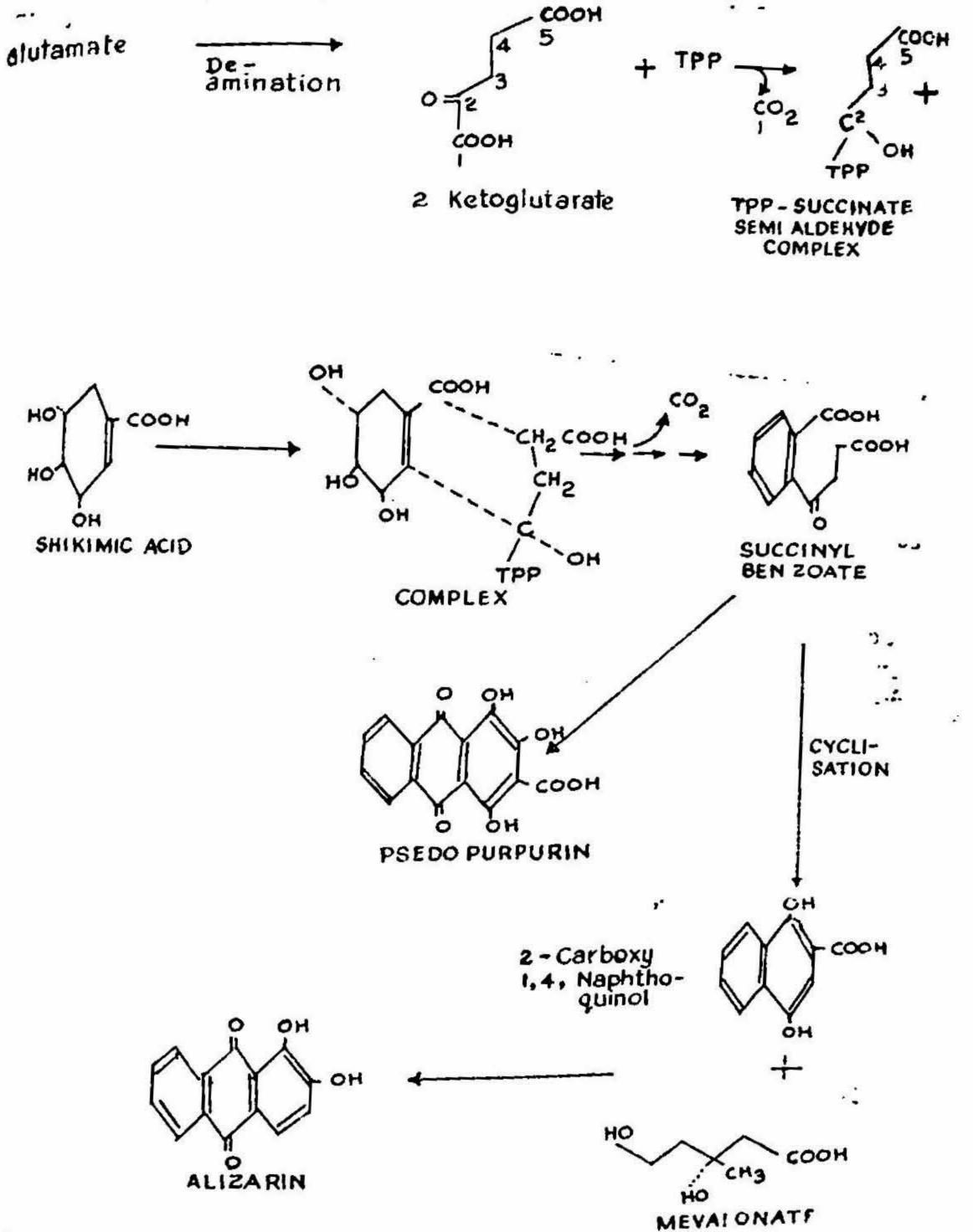


Fig. 3. Anthraquinone biosynthesis from shikimic acid.

One year old Rubia tinctorum plants fed with 2-14C mevalonate is found to incorporate it into 4-anthraquinone pigments, the highest amount of incorporation being into ubiadin and pseudopurpurin. Decarboxylation of pseudopurpurin and degradation of purpurin and other anthraquinones showed that mevalonate is incorporated into

ring C of the anthraquinone nucleus. The C₂ of mevalonic acid provided the side chain carbon atom and also the ring carbon atom at $C_4^{65,66}$. A new intermediate, synthetic 14C-O-succinyl benzoic acid has been found to be efficiently incorporated into naphtho. quinones and anthraquinones of *Rubia peregrina*⁶⁷.

6. Pharmaceutical studies

The cathartic action is attributed to the presence of anthraquinones or related compounds. The sennosides A and B do not account for the total purgative action of senna⁴⁰. Urumaco, the fresh flowers of *Adipera fahnei*, used in the Andes as a purgative, do not contain any sennosides⁶⁹. But it contains numerous anthraquinone derivatives including anthraquinols. It is reported that the content of non-rhein glycosides is 10-15% of the total glycosidal content of senna leaf and 2-5% of pods and that the non-rhein glycosides are as active as the sennosides⁶⁸. Khorona and Sanghavi⁷⁰ isolated glycosides of rhein and chrysophanic acid from the pods of *C. acutifolia by* acidification of aqueous extract to pH 3.0. A mixture of these two glycosides is more active biologically than either alone. In view of the fact, that the non-rhein glycosides are equally active cathartically as the sennosides, the study made to determine quantitatively the different classes of anthracene derivatives of senna may be of interest⁷¹. Determination from four official samples of senna gave the following average values : anthracene compounds $2 \cdot 24-3 \cdot 06\%$; total anthraquinones $0 \cdot 97-1 \cdot 11\%$; and anthranols $1 \cdot 23-1 \cdot 97\%$

The relative purgative action of twelve 1,8-dihydroxy anthracene derivatives including free anthraquinone, anthrone and dianthrone forms, arthraquinone-O-glycosides and dianthrone-O-glycosides are compared with senna powder using the production of wet faeces by mice as a criterion of purgation ⁷². Rhein dianthrone had higher purgative activity. Sennidin had higher activity than reported previously. Rhein-anthrone, rhein, aloe-emodin and chrysophanol are less active. Emodin is inactive in mice.

The preparations of senna act on the intestine by exciting peristalsis without affecting the functions of stomach and duodenum. It is painless in use and the intestine does not become habituated to their use⁷³. The purgative action depends on the amount of free hydroxy anthraquinone and the ease of decomposition of the corresponding glycosides. An infusion of dried senna is activated in the large intestine only, possibly by colon bacillus⁷⁴. The activation of colon bacillus appears to proceed by hydrolytic cleavage of the anthraquinone glycosides. The aglycones are then activated in the colon⁷³. The activated anthraquinone derivatives act directly on the smooth muscles of colon⁷⁴. As a result of which the peristaltic movements of the colon are stimulated and increased. Also the absorption of water is decreased and a softer faecal mass is formed⁷⁷. Auger *et al.*⁷⁸ reported that 1,8-dihydroxy anthraquinone monoglycoside had superior colonic propulsive activity than the aglycon when given orally to mice, resulting in greater purgative activity. They conclude that anthraquinone derivatives act on the colon after absorption by the small intestine. The anthraquinone derivatives are more favourably absorbed in the form of glycosides and also probably glucose moiety protects the active principle against degradation.

Therapeutical doses stimulate intestinal peristalsis, the aperient effect ensuring in about 7 to 12 hours. It may be given to children and elderly persons when a tolerably active purge is required and it is good to combine a saline aperient such as epsom salt with it⁷⁹.

7. Medicinal preparations

Senna pods are usually administered in the form of an infusion, four to twelve pods being soaked in about five fluid ounces of warm water for about 12 hours⁷³. Leaf could also be treated similarly. The preparation of a senna liquid extract is described in B.P.C.⁸⁰. A liquorised powder compound is also prepared from senna leaf. Liquid preparations from senna are not fully active and are less stable². A chemically and biologically standardised powder of the pericarp has been prepared which remains stable for a long time². This powder is used in the preparation of granules and is found to contain the whole of the theoretical laxative activity. Seventy per cent ethanol extracts of senna were found to contain 85–90% of the sennosides. The extract is stable in amber coloured bottles for 96 hours and is used for the preparation of senna tablets and syrup⁸¹.

A few other medically useful preparations involving senna extract or constituents of other plants of *Cassia* have been described. Laxative-hypnotic drug⁸² combinations which allow a normal sleeping rhythm and are useful in the prophylaxis of the drug abuse (suicide by sleeping tablets) consist of aloe-extract, phenyl ethylbarbituric acid and optional synergists, *e.g.*, Frangula extract, senna extract or diethylbarbituric acid. Medicaments useful for smoothening digestive mucous tissues contain as active ingredients, mucilage from seeds of *Cassia absus*⁸³. A preparation obtained by stirring molten tin with 25% each of powdered tamarind, *Ficus religiosa* and *C. auriculata* capable of stimulating appetite and digestion and rejuvenating the body system has been described⁸⁴. Externally, powdered leaves mixed with vinegar and made into a plaster are applied locally in certain skin diseases.

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References

- In The Chemistry of Drugs, Ernest Benn Limited, London, 1. EVER, N. AND CALDWELL, D.
- RYAN, H. A. Pharm. J, 1951, 167, 115. 2. Trop. Sci., 1962, 4, 159.

.

- Proceedings, First All India Workshop in Medicinal and Aron Plants, Bangalore, 1973.
 - J. Am. Pharm. Asso, 1951, 40, 52.
 - Rev. Brasil. Farm., 1956, 37, 119. Farmakal i toksikol, 1941, 4, 30.
 - Helv. Chim. Acta, 1949, 32, 1892.
 - Pharm. Tijdschr. Belg., 1962, 39, 67.
 - Helv. Chim. Acta, 1965, 48, 1911.
 - Pharm. Acta. Helv., 1967, 42, 37.

GROTE, I. W. AND 5. WOODS, M.

3. SEAFORTH, C. E.

- 6. LUCAS, V.
- 7. Moszov, I. E.
- 8. STOLL, A., BECKER, B. AND KUSSMAUL
- 9. LEMLI, J.
- 10. SCHMID, W. AND ANGLIKER, E.
- 11. LEMLI, J. AND CUVEELE, J.

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- 12. LEMLI, J., DEQUEKER, R. AND CUVEELE, J.
- 13. FAIRBAIRN, J. W. AND SHRESTHA, A. B.
- 14. CRELLIN, J. K., FAIRBAIRN, J. W., FRIEDMANN, C. A. AND RYAN, H. A.
- 15. ROMANOVA, A. S., SEMAKINA, N. D. AND BENKOVASKII, A. I.
- 16. ATAL, C. K. AND KAPUR, B. M.

٠

- 17. THAROCILLO, A NEUBERN DE TOLEDO
- 18. HAUPTMANN, H. AND NAZARIO, L.

- Pharm. Weekblad, 1964, 99, 589.
- J. Pharm. Pharmacol., 1966, 18, 467.
- J. Pharm. Pharmacol., 1961, 13, 639.

Rast. Resur., 1968, 4, 342.

In Cultivation and Utilisation of Medicinal and Aromatic Pla Regional Research Laboratory, Jammu-Tawi, 1977. Anais, Fac. Farm e odontol Univ. Sao Paulo, 1949, 7, 105.

J. Am. Chem. Soc., 1950, 72, 1492.

CHEMISTRY OF SENNA

Planta. Med., 1973, 23, 368.

- 19. VARSHINEY, S. C., Rizvi, S. A. I. and Gupta, P. C.
- 20. MATOS, F. J. A.
- 21. JOSE, G. C. AND CAIROLI, J. E.
- 22. KAPADIA, G. J. AND KHORANA, M. L.
- 23. SACCO, T.
- 24. KHALEQUE, A. H. AND NESA, S.
- 25. KAJI, N. N. Khorana, M. L. and Sanghavi, M. M.
- 26. KARIM, M. A.
- 27. RICHARD, W.
- 28. KUMAR, A., PANDE, C. S. AND KAUL, R. K.
- 29. JAIN, S. K.
- 30. JAEGAR, P., ANTON, R.

- Anais. Fac. Farm. e odontol Univ. Sao Paulo, 1957-58, 15, 129. Ind. Y. Quim., 1960, 20, 141.
- Lloydia, 1962, 25, 55.

Riv. agr. sub-trop e trop, 1958, 52, 119. Science Res. (Decca), 1970, 7, 63.

- Indian J. Pharm., 1968, 30, 8.
- Pakistan J. Sci. Res., 1952, 2, 114. Bol. Acad. Nacl., Farm, 1942, 4, 133.
- S. Indian J. Chem., 1966, 4, 460.
 - In Medicinal Plants, National Book Trust, India, 1968. Plant. Med. Phytother, 1969, 3, 204.

. . . .

- AND DUQUENOIS, P.
- 31. THARCILLO, A. NEUBERN DE TOLEDO AND GROTT, A. S.
- 32. ANTON, R. AND DUQUENOIS, P.
- 33. SUBRAMANIAN, S. S. AND NAGARAJAN, S.
- 34. SUBRAMANIAN, S. S. AND NAGARAJAN, S.
- 35. DUQUENOIS, P. AND ANTON, R.
- 36. RIZVI, S. A. T., GUPTA, P. G. AND KAUL, R. K.

Anais. Fac. Farm e odontol Univ. Sao Paulo, 1951, 9, 179.

Indian J. Pharm., 1969, 31, 110.

Indian J. Pharm., 1970, 32, 70.

Ann. Pharm. Fr., 1968, 26, 607.

Planta Med., 1971, 19, 222.

- 37. SABER, A. H., BALBAA, S. I. AND ALBERT, T. A.
- 38. TAKIDO, M. AND TAKAHASHI, S.
- 39. NYDIA, M. K.
- 40. ANTON, R. AND DUQUENOIS, P.
- 41. SHAH, C. S. AND SHINDE, M. V.
- 42. NAZIROV, Z. H.
- 43. PARIS, R. AND CHARTIER, J.
- 44. MARJORIE, A.
- MESMER, W. M., FARNSWORTH, N. R., PERSINOS, G. J. AND WILKES, J. D.
- 46. BALANSARD, J. AND

•

- Lloydia, 1962, 25, 238.
- Syoyakugaku Zasshi, 1963, 17, 43.
- J. Am. Pharm. Assoc., 1957, 46, 271.
- Ann. Pharm. Fr., 1968, 26, 673.
- Indian J. Pharm., 1969, 31, 27.
- Trudy Tashkent Farmatsevt Inst., 1960, 2, 173. Ann. Pharm. Fran., 1948, 6, 30.
- J. Biol. Chem., 1949, 177, 169.
- J. Pharm. Sci., 1968, 57, 1996.

Ann. Musee. Colonial Murseille, 1940 (5), 8, Fase 2, 1-16.

- MOREL, A.
- 47. DEQUENOIS, P. AND ANTON, R.
- 48. CHATTERJEE, A. AND BHATTACHARJEE, S. R.
- 49. MARGARIDA, F. A. AND CORREIA, A. A.
- 50. POETHKE, W., Anand Rao, D and Loescher, K. L. D.
- 51. TAKODO, M., TOSHIO, N. AND NILTA, K.
- 52. RAKHIMOVA, A. K.
- 53. SABER, H. A., BALBAA, S. I. AND AWAD, A. T.

- Planta Med., 1968, 16, 184.
- Bull. Nat. Inst. Sci., India, 1965, 31, 141.
- Gericia de orta, 1968, 16, 193.
- Pharm. Zentralh, 1968, 107, 571.
- Nippon Daigaku Yakugako Kenkyu Hokoku, 1960, 4, 18.
- Doklady Akad. Nauk Agerbaidzhan, S.S.R., 1955, 11, 295. Bull. Fac. Pharm., 1961–62, 1, 7.

54. SELVARAJ, Y. AND SUBHAS CHANDER, M.

•

55. FAIRBAIRN, J. W. AND SHRESHTHA, A. B.

56. ANTON, R. AND DUQUENOIS, P.

57. SAUL PATAI

58. LEISTNER, E. AND ZENK, M. H.

59. LEISTNER, E.

60. FAIRBAIRN, J. W. AND MUHTADI, F. J.

61. LEISTNER, E. AND ZENK, M. H.

62. CAMPBELL, I. M.

63. LE DUC, M. M., DANSETTE, P. M. AND AZERAD, R. G.

64. LEISTNER, E. AND ZENK, M. H.

In Annual Report, Indian Institute of Horticultural Research, 1973.

Phytochemistry, 1967, 6, 1203.

C.R. Acad. Sci. Paris, Ser. D., 1968, 267, 1227.

In The Chemistry of Quinonoid Compounds, Part 2, John Wiley and Sons, New York, 1974.

Chemical Communications, 1969, 5, 210-211.

Phytochemistry, 1971, 10, 3015. Phytochemistry, 1972, 11, 215.

Tetrahedron Letters, 1967, No. 5, p. 475.

Tetrahedron Letters, 1969, No 54, p. 4777. Eur. J. Biochem., 1970, 15, 428.

Tetrahedron Letters, 1968, No. 7, p. 861.

Chem. Comm., 1967, 21, 1125-1126.

:

- BURNETT, A. R. AND 65. THOMSON, R. H.
- 66. BURNETT, A. R. AND THOMSON, R. H.
- 67. DANSETTE, P. AND AZERAD, R.
- 68. FAIRBAIRN, J. W. AND SALEH, M. R. I.
- 69. CARL, S. AND TERON, L.R.

٠

- 70. KHORANA, M. L. AND SANGHAVI, M. M. t
- 71. NIHOUL, L.
- 72. FAIRBAIRN, J. W. AND Moss, M. J. R.

- J. Chem. Soc. (C), 1968, p. 2437. Biochem. Biophys. Res. Comm., 1970, 40, 1090. ų. J. Pharm. Pharmacol., 1951, 3, 918. Fac. Farm. Univ. Andes (Venezeula), 1958, 1, 7.
- J. Pharm. Sci., 1964, 53, 110.
- J. Pharm. (Belg.), 1960, 15, 49. J. Pharm. Pharmacol., 1970, 22, 584.

I.I.Sc.-8

400	
10%	
370	

- 73. CARAVAGGI, A. AND MANFREDI, A.
- 74. OKADA, T.
- 75. SCHMIDT, L.
- 76. VALETTE, G. AND MARIE, L. H.
- 77. HARDCASTLE, J. D. AND WILKINS, J. L.
- 78. AUGER, J., Bretaudeau, J., Niviere, R., Foussard, B. and Cluzan, R.
- 79. NADAKARNI, A. K.

80.

Bidl. Chim. Farm., 1937, 76, 117.

Tohoku J. Expl. Med., 1940, 38, 33.

Naunyn. Schmiedeberas Archiv. Expt. Pathol. Pharmakol., 1955, 226, 207.

1

Therapie, 1957, 12, 885.

Gut, 1970, 11, 1038.

1

1

Ann. Pharm. Fr., 1975, 33, 569.

In Indian Materia Medica, Popular Prakashan Private Ltd., Bombay, 1976.

Department of Pharmaceutical Sciences. In British Pharmaceutical Codex. The Pharmaceutical Press, London, 1968.

- 81. GUPTA, G. K., RAU, P. R. Res. Ind., 1975, 20, 124. AND VISHWAPAUL
- 82. GUNTER, H. Ger. Offen, 1973, 2, 128, 447 (Patent).
- 83. LABORATORIES, H. AND FAURE, S. A.
- 84. VASANTH, S., BHARATHI, R. B., PURUSHOTHAMAN, K. H. AND NARAYANASWAMI, V.
- 85. RAI, P. P.

- Fr. Demande, 1974, 2, 167, 219 (Patent).
- J. Res. Indian Med., 1971, 6, 216.
- Curr. Sci., 1977, 46, 814.