J Indian Inst Sci., May-June 1994, 74, 411-471. © Indian Institute of Science.

# Cobalt carbonyls : A versatile reagent and catalyst in organic synthesis

JAVED IQBAL\*, BEENA BHATIA AND VIBHA KHANNA Department of Chemistry, Indian Institute of Technology, Kanpur 208 016, India.

Received on March 2, 1994, Revised on June 20, 1994

#### Abstract

Cobalt carbonyls are versatile reagents for the insertion of carbon monoxide between carbon-cobalt bond. This methodology has been utilized for the synthesis of some naturally occurring compounds. In this review, cobalt carbonyl-mediated novel reactions (2+2+1 and 2+2+2 cycloaddition reactions, Nicholas reaction, carbonylation reactions, ec) have been discussed. Mechanisms of these reactions have also been presented.

Key words : Dicobalt octacarbonyl, cycloaddition, propargylic cobalt complex, carbonylation.

# Introduction

One of the outstanding features in the growth of contemporary organic chemistry has been the emergence of transition metal-mediated organic reactions at the frontiers of organic synthesis. Its instant acceptance by synthetic chemists is primarily due to the fact that transition metal-mediated organic transformations are extremely versatile and experimentally convenient. Recent development in the catalysis of organic reaction by transition metals has paved the way for achieving a remarkable level of chemo- and stereoselectivity, and for certain reactions this achievement has reached an extent where near enzyme-like selectivity has been witnessed. This development has clearly expanded the arsenal of synthetic chemist which has eventually facilitated the intensity of assault on some of the most challenging problems of synthesis.

Transition metal carbonyls<sup>1-2</sup>, particularly from Fe, Co, Rh, Ni, Pd, play an important role in industrial chemistry<sup>3</sup>, since they allow for high selectivity and economic efficiency in such processes as hydrogenation, hydroformylation<sup>4</sup>, oxidation<sup>5</sup>, epoxidation<sup>6</sup>, etc. In addition to this, metal carbonyls<sup>7-12</sup> derived from Mo, Cr, W have also been successfully applied to a wide range of unique organic transformations. The introduction of one carbon in organic substrates under the aegis of metal carbonyls constitutes a very important transformation in contemporary organic synthesis<sup>13</sup>. Among the various transition metal carbonyls, the carbonyls derived from Co have made outstanding contribution towards achieving a wide range of organic transformations, like hydroformylation, carbonylation, oxysilylation,

<sup>\*</sup> For correspondence.

cycloaddition reaction of alkynes, Nicholas reaction, etc. Impressive advances have been made in the domain of cobalt carbonyl-mediated organic synthesis over the last one decade as clearly evident from the remarkable level of efficiency and selectivity achieved during the synthesis of complex natural products.

In view of the importance of these reactions this review covers literature on cobalt carbonyl-catalysed or mediated reactions in organic synthesis. The review has been divided into the following sections.

- 1. Cycloaddition reactions
  - 1.1 [2+2+1] Cycloaddition reactions

1.2 [2+2+2] Cycloaddition reactions

- 2. Nicholas reaction
- 3. Carbonylation reaction
- 4. Miscellaneous reactions

# 1. Cycloaddition reactions

# 1.1. [2+2+1] Cycloaddition reactions

[2+2+1] Cycloaddition reactions (Pauson-Khand reaction) is a novel and useful method for the synthesis of cyclopentenone derivatives. This reaction, first reported by Pauson and Khand<sup>14</sup> in 1973, involves the cocyclization of alkynes with alkene and carbon monoxide under the aegis of dicobaltoctacarbonyl. This transformation is a [2+2+1] cycloaddition which involves thermally stable hexacarbonyldicobalt complex 1<sup>15</sup>, obtained by co-ordination of Co<sub>2</sub>(CO)<sub>8</sub> with alkynes in hydrocarbon solvents of ether. Subsequent reaction of complex 1 with alkene followed by insertion of carbon monoxide leads to the formation of cyclopentenone (eqn 1).



This reaction is compatible with a wide range of functionalities like ether, alcohols, tertamines, thioethers, ketones, ketals, esters, *tert*-amides<sup>16</sup> and aromatic rings including benzene, furan and thiophene<sup>17</sup>.

This reaction can be divided into two categories: (a) Intermolecular reaction, and (b) Intramolecular reaction.

# 1.1.1. Intermolecular [2+2+1] cycloaddition reactions

Intermolecular reaction of strained alkenes<sup>18</sup> with acetylene and Co<sub>2</sub>(CO)<sub>8</sub> reacts with norbornene and its derivatives to generate cyclopentenone derivatives 3 (eqn 2). Similarly,

unsymmetrical alkynes react with 4 via its Co-complex 5 to provide 8-oxabicyclo [3.2.1] oct-6-ene-3-one 6 in quantitative yields (eqn 3).



Nitrogen-bridged bicyclic<sup>19</sup> systems have also been shown to undergo Pauson-Khand cycloaddition. Thus, 7 undergoes addition to acetylenic Co-complex 2 to provide 8 in good yields (eqn 4). High regioselectivity in incorporation of the unsymmetrical alkyne in the product is a characteristic feature of this reaction. In addition, these reactions also occur with high stereoselectivity as exo-adduct is obtained as the predominant product.



Simple unstrained alkenes<sup>20</sup> are unreactive under these reaction conditions; however, alkenes containing electron-withdrawing groups **9** react to yield conjugated dienes<sup>21</sup> (eqn 5). Alkenes containing one or more electron-withdrawing groups react with acetylenic cobalt complex<sup>1</sup> to give conjugated diene. However, due to the low yields, this reaction cannot be used as a general synthetic route to conjugated dienes.

These reactions are believed to occur via the insertion of alkene into the cobalt-acetylene  $complex^{22}$  to give 2a which subsequently incorporates the carbon monoxide to afford 2b which on reductive elimination of cobalt leads to cyclopentenone(Scheme 1). The formation of diene may be occurring via a similar pathway involving the hydrogen migration followed by a process of reductive elimination (Scheme 2).



SCHEME 2.

Styrene and substituted styrene<sup>23</sup> represent the borderline cases where both modes of reaction (*i.e.*, cyclopentenone and diene formation) are observed (eqn 6).



Usually terminal alkenes<sup>24</sup> give poor yields and regioselectivity. Krafft<sup>25</sup> has shown that alkenes containing groups capable of acting as soft ligands at a homoallylic position give both

enhanced yields and regioselectivity. This may be a result of coordination of heteroatom prior to insertion (Scheme 3).



SCHEME 3.

Schore<sup>26</sup> has synthesised 4,5-disubstituted 2-cyclopentenones 11 from the cycloaddition product 10 of norbornadiene using cuprate addition followed by retro Diels-Alder reaction (Scheme 4).



**SCHEME** 4

This reaction is regio- and stereoselective<sup>27</sup> as with bicyclic alkene 12; the less-hindered face of the  $\pi$ -bond preferentially reacts to give exo-ring fusion product 13 exclusively (eqn 7).



This high selectivity has been exploited by Schore and co-workers during the synthesis of guaianolide and pseudoguaianolide<sup>28</sup>. The stereochemistry of the ring fusion as present in the key intermediate **15** is remarkably achieved in the first step using the Pauson-Khand reaction on alkene **14** (Scheme 5).



SCHEME 5.

Interestingly, cyclopropane  $ring^{29}$  is tolerated in the acetylenic partner during the cycloaddition on cyclic alkenes to give 17. This methodology has been used during the synthesis of linearly fused triquinanes 18 from cyclopropyl acetylene 16 and cyclopentene (Scheme 6).



SCHEME 6.

Serratosa and co-workers<sup>30</sup> have synthesised angularly fused triquinanes 21 starting from 19 and 20 (eqn 8).

18(39)



Smit-Caple and co-workers<sup>31</sup> have shown that intramolecular Khand reaction could be carried out with an increased efficiency in a solvent-free system with the substrate adsorbed on the surface of chromatography adsorbent (dry state adsorption conditions). Intermolecular reaction between 22 and 23 was conducted by conventional Pauson-Khand reaction. Later, they have shown that under dry state adsorption conditions the [2+2+1] cycloaddition of 24 with 5 proceeds quite smoothly in high yields (Scheme 7). These reactions have been shown to occur on strained alkene using Al<sub>2</sub>O<sub>3</sub> or MgOSiO<sub>2</sub> as dry media.



SCHEME 7.

Recently, Krafft and co-workers<sup>32</sup> have shown that the regioselectivity of co-cyclization is directed by the use of soft atom like sulfur or nitrogen. They observed that alkene containing S or N at homoallylic position is more effective in controlling the regioselectivity as compared to alkene containing S or N at allylic or homoallylic position (Scheme 8).



SCHEME 8.

JAVED IQBAL et al.

Jeong and co-workers<sup>33</sup> have devised a highly efficient one-pot strategy for the preparation of aza-bicyclic compounds vu Nicholas reaction (see Section 2) with amidic nitrogen nucleophiles followed by Pauson-Khand reaction (Scheme 9).



SCHEME 9

# J.J.2. Intramolecular 2+2+2 cycloaddition reactions

Intramolecular Pauson–Khand reaction was first reported by Schore and Croudace<sup>34</sup> in 1981. This methodology has been used during the synthesis of various natural products. Enynes cyclize, upon complexation to Co<sub>2</sub>(CO)<sub>8</sub> and subsequent heating, to give bicyclic enones. The most extensively studied is the synthesis of bicyclo [3.3.0] oct-1-ene-3-one from hept-1-cne-6-yne (eqn 9). Hex-1-en-5-yne produces a mixture of products of trimerization of the alkyne functionality.



The presence of bicyclic [3.3.0] octane ring system in a variety of biologically active natural products has generated considerable interest in the synthesis of its functionalized derivatives. In these cycloadditions, substitution on both the alkyne as well as the chain linking the alkyne and the alkene is often readily tolerated.

Hua and co-workers<sup>35</sup> have prepared the key precursor 26 for the synthesis of optically active pentalene and racemic pentalenolactone E methyl ester from enyne 25 (eqn 10).



Later. Seto et  $al^{26}$  have shown that  $Co_2(CO)_8$ -mediated cyclization of an acyclic-enyne 27 provides bicyclic pentenone 28, which can be elaborated to antibiotic pentalenolactone G 29 (Scheme 10).

Magnus and co-workers<sup>37</sup> have exploited intramolecular Pauson-Khand reaction to the total synthesis of coriolin **32**, a linearly fused triquinane. The key intermediate **31** was prepared in one step from the readily available enyne **30** (Scheme 11).



SCHEME 10.



SCHEME 11.

Magnus has systematically examined the factors that contribute to the stereoselectivity shown in intramolecular Pauson-Khand reaction. This methodology has been used for the synthesis of hirsutic acid<sup>38</sup> 35 from enyne 33 via bicyclopentenone 34 (Scheme 12).



SCHEME 12.

Magnus<sup>39</sup> has developed an elegant approach to the synthesis of quadrone 38 from the key precursor bicyclo [3.3.0] oct-1-ene-3-one 37 prepared in one step from the enyne 36 in a highly stereoselective manner (Scheme 13). O



SCHEME 13.

Magnus and co-workers have also achieved a stereoselective synthesis of a carbocycline analogue<sup>40</sup> **41** using the enyne **39** via the cyclopentenone **40** (Scheme 14).

The presence of alkene in a ring (e.g., 42) is compatible with intramolecular cyclization as angularly fused triquinanes like bisnorisocomene<sup>41</sup> can be synthesised from cyclopentenone



SCHEME 14

43 stereoselectively. This reaction has the limitation as only trisubstituted alkenes and simple terminal alkynes can be used for the cyclopentenone formation (eqn 11).



A stereocontrolled approach to pentalenes<sup>42</sup> has been shown by using the above methodology (eqn 12).



Serratosa and co-workers have developed an exceptionally efficient approach to triquinacenes<sup>43</sup> **45** making use of similar intramolecular cycloadditions of cyclic alkenes **44** containing alkynyl substitution (eqn 13).

Billington and co-workers<sup>44</sup> have cyclized substituted ally1-propargyl ethers **46** to give 3oxa bicyclo [3.3.0] oct-5-en-7-ones **48** via hexacarbonyl dicobalt complexes **47**. Hydrogenation



of 48 followed by deprotection afforded the key intermediate<sup>49</sup> for the synthesis of tetrahydro anhydroaucubigenone 50 (Scheme 15).



SCHEME 15.

Schreiber<sup>45</sup> and Smit<sup>46</sup> have synthesised polyheterocycles **52** and **54** by combining Nicholas and Pauson-Khand cycloaddition reactions using enyne ether **51** and **53**, respectively (Scheme 16).



SCHEME 16.

Later, Smit and co-workers<sup>47</sup> have shown unusual effect on the efficiency of Co-mediated conversion of an enyne-ether 55 into the corresponding bicyclo[3.3.0] octenone 56 by adsorption of Co-complexed enyne-ether on to silica gel under  $O_2$  or air (Scheme 17).



SCHEME 17.

Veretenov and co-workers<sup>48</sup> have developed a simple route for the synthesis of polycyclic inearly and/or angularly fused compounds **58** from **57**. This cycloaddition occurs with participation of double bond, having an electron-withdrawing group (eqn 14).



Schreiber and co-workers<sup>49</sup> have developed an efficient method which provides a milder und more stereoselective alternative to the corresponding thermal reactions. Tertiary amine ixide (e.g., N-methylmorpholine-N-oxide, NMO) readily promotes intramolecular Pauson-Chand cyclization at room temperature on **59** under an inert atmosphere. Due to the milder condition required, this reaction tolerates various functional groups like alcohols, silvl ether thers, acetals, remote olefins, etc., and leads to the formation of **60** and **61**. One of the sutstanding features of this reaction is the high level of stereoselectivity as compared with iltrasonic or thermal reaction (Scheme 18).



A novel route to the precursor **63** of  $(\pm)$  Loganine **64** has been developed by Jeong and co-workers<sup>50</sup> from homoally1-propargy1 acetal **62** (Scheme 19).



SCHEME 19

Recently, Hoye and Suriano<sup>51</sup> have shown that electron-deficient alkynones can be interor intramolecularly cyclized to give bicyclic enediones in good yields. They have observed that there is a remarkable effect on the reactivity by changing the solvent. The effect of solvent is evident from the reaction of **65** in acetonitrile with norbornene which gives **66** by intermolecular addition whereas the enyne **67** on intramolecular cyclization affords **68** m high yields. A similar transformation in methanol was quite sluggish (Scheme 20).



SCHEME 20.

Interestingly, the highly functionalized alkyne 69 undergoes intramolecular cyclization in the presence of NMO to afford 70 in high yields and good stereoselectivity<sup>49</sup> (eqn 15).



Smit and co-workers<sup>52</sup> have synthesised several fenestrane derivatives **72** based on intramolecular Pauson-Khand reaction followed by [2+2] photocycloaddition on intermediate **71** (eqn. 16).



Krafft and co-workers<sup>53</sup> have shown the rate of the thermal intramolecular Pauson-Khand cycloaddition can be enhanced by 1,6-enyne-bearing co-ordinating ligands (sulfur or oxygen atom) in the homo and bishomopropargylic position. They have shown that sulfur provides more acceleration than oxygen (Scheme 21).



SCHEME 21.

# 1.2. [2+2+2] Cycloaddition reactions

The discovery of new synthetic methods has already made possible to construct the most complex natural products and the most 'unnatural' assemblies. Despite these advancements

## COBALT CARBONYLS

there remains much room for improvement of synthetic strategies to get the chemo-, regioand stereoselectivity of the compounds. A simple analysis showed that a more powerful strategy would be based upon the [2+2+2] cycloadditions of the unsaturated moieties. CpCo(CO)<sub>2</sub> as a catalyst was found to promote the successful execution of [2+2+2] cycloadditions<sup>54</sup>. The many previously unattainable molecules generated in this way have been used as a starting material for the preparation of several unnatural and natural products of theoretical, medicinal and synthetic interest. Two decades earlier, it was found that CpCo(CO)<sub>2</sub> catalyses a variety of [2+2+2] cycloadditions involving  $\alpha_i$ , $\alpha_i$ -diynes to give annelated benzenes<sup>55</sup>. In order to get chemoselectivity, bulky alkynes such as trimethylsilylalkynes were employed<sup>36,57</sup> (eqn 17). Cobalt-catalysed cocyclization reaction was used in silicon-directed intermolecular regioselective Friedel–Crafts acylation<sup>57</sup> (eqn 18).



To understand the mechanism of these reactions<sup>58-60</sup> many studies have been carried out which resulted in the isolation of two intermediates **73** and **74**. Cyclobutadiene complexes derived from both **73** and **74** are obtained as byproducts in catalytic reactions employing  $\alpha$ ,  $\omega$ -diynes<sup>57</sup> and are responsible for some of the catalyst depletion since they appear to be unsuitable as precursors for any catalytic intermediates<sup>61</sup>.



The trimethylsilyl group which is used extensively for controlling the chemo- and regioselectivity has a pronounced tendency to promote  $\alpha$ -selectivity in the metallacycle. This effect was synthetically demonstrated in the formation of 77 as the sole isomer on cocyclization of 1-trimethylsily 1-1, 5-hexadiyne 75 and trimethylsilylacetylene<sup>53</sup> 76 (Scheme 22). On the other hand, if more Me<sub>3</sub>S1 groups are present, *i.e.*, 78 then the reaction proceeds



SCHEME 22

via o-xylylene formation and intramolecular ring closure to give benzhydrindane nucleus<sup>62</sup> 79 (Scheme 23).



# SCHEME 23

The CpCo(CO)<sub>2</sub>- catalysed [2+2+2] cycloaddition of three alkyne units was applied to total synthesis of a variety of natural products such as antitumor anthracyclene aglycones<sup>63</sup> (Scheme 24) and the protoberberine alkaloids<sup>64</sup>. The protoberberine **81** is readily prepared by cocyclication of **80** with bis(trimethylsilyl) acetylene<sup>65</sup> (Scheme 25).



1,2-Dihydrocyclobutabenzenes are used in the construction of a host of theoretically interesting benzenoid hydrocarbons<sup>66-68</sup>. Initially, 1,5-hexadiyne undergoes one-step trimerization which on oxidative photocyclization gives the two isomeric dicyclobutaphenanthrenes showing the tandem cyclization–cycloaddition reaction<sup>69</sup> (Scheme 26). A novel series of



## Scheme 26

compounds called as multiphenylenes have been prepared using cobalt complexes as catalyst<sup>70,71</sup> (Scheme 27). Apart from their use in the synthesis of different strained ring systems, the ,2-dihydrocyclobutabenzenes have been used in producing polycyclic systems<sup>72,73</sup>.



# Scheme 27.

Enediyne 82 undergoes intramolecular cyclization to yield stereospecific cyclohexadiene complex 83 in the presence of stoichiometric amount of CpCo(CO)<sup>74,75</sup> (eqn 19). A sequence of 2D NMR experiments in conjunction with labelling experiments has shown the presence of intermediates which on rearrangement gives the product.



In the same way, enediynes **84** with internal double bonds undergo intramolecular cyclization<sup>76</sup> to give **85**. This cyclization procedure proceeds efficiently and with remarkable stereoselectivity, both with respect to the stereochemistry of the original double bond and of cobalt<sup>77</sup> (eqn 20). CoCp



This reaction showed that the steric encumberance of the double bond has little influence on the success of the reaction. This advantage was utilized in the preparation of tricyclic diene 87 from a substrate containing tetrasubstituted double bond<sup>78</sup> 86 (eqn 21).



Intermolecular [2+2+2] cycloadditions of enynes 88 on co-oligomerization with BTMSA gave mainly cyclobutadiene 89 rather than expected bicycle<sup>60,79</sup> 90 (Scheme 28).



428

Based on the 'Tandem principle' described earlier, CpCo(CO)<sub>2</sub>-catalysed synthesis of steroids was achieved starting from 1.5 hexadiyne. Alkylated 1.5-hexadiyne 91 underwent tandem cocatalysed cyclization followed by intramolecular ring closure via  $\sigma$ -xylene formation to give key precursor 92 of (±) estrone<sup>80</sup> (Scheme 29).



## SCHEME 29.

Another way<sup>81</sup> of synthesising steroids is using  $CpCo(CO)_2$  in which the -BCD portion of their framework would be fused to a pre-existing aromatic A-ring<sup>82,83</sup> (Scheme 30). A



SCHEME 30.

diastercoselective synthesis of steriod<sup>84</sup> has been achieved using the enediyne **93**. Cyclization followed by demetallation under acidic conditions gave the known estrapentaneol<sup>85</sup> **94** (Scheme 31). Another approach to the steroid synthesis employing CpCo(CO)<sub>2</sub> as a matrix is O->ABCD, *i.e.*, all four rings are assembled in one step from enetriyne **95** to give B-ring aromatic derivatives with the complete control of the crucial stereochemistry of the C,D-ring juncture<sup>86</sup> (Scheme 32).



SCHEME 32.

Highly crowded steroids<sup>87</sup> were prepared using the enediyne 96a to give 97 whereas the corresponding silylated derivative 96b afforded a highly stereoselective formation of 98 (Scheme 33). The outcome of this reaction demonstrates once again the unique ability of the catalyst to make highly hindered compounds.



SCHEME 33.

In the early 1970s, several groups independently discovered that cobalt complexes could cocyclize alkynes with nitriles to furnish pyridine in stoichiometric and catalytic reactions<sup>88,89</sup> (eqn 22).



This reaction can be used in producing some very rare isoquino[2,1-b]-2,6-naphthyridine nucleus<sup>64</sup> 99. Similarly, the 2-azaanthracene 100 framework can be obtained efficiently (Scheme 34).



SCHEME 34.

Cocyclization of bis (trimethylsilyl) **101a** or bis (trimethylstannyl) di-2-propynyl ether **101b** with acetonitrile provides a synthetic route to 1,3-dihydro-6-methyl-4,7 bis (trimethylsilyl) **102a** or bis (trimethylstannyl)-furo[3,4-c] pyridines **102b**. This methodology has been used for the total synthesis of Vitamin  $B_6$  **103**<sup>90</sup> (Scheme 35).



SCHEME 35.

Employment of isocynates<sup>91,92</sup> in place of nitrile on cocyclization with alkynes afforded polyheterocyclic systems (eqn 23). In simple cocyclizations leading to substituted pyridone,



regioselectivity was not observed, whereas when bulky substituent at the  $\alpha$ -position of  $\omega$ -alkynyl isocyanates was present (*e.g.*, trimethylsilyl) good chemo- and regioselectivity were observed. Application of the above methodology has led to formal synthesis of the antitumor alkaloid camptothecin<sup>92,94</sup> **104** (Scheme 36). Similarly, incorporation of the 6-heptynenitrile **105** unit into the indole gives the basic skeleton of the ergot alkaloids<sup>85</sup> (eqn. 24).





It has been found that trimethylsilylalkynes undergo [2+2+2] cycloadditions under lowtemperature photolytic conditions in the presence of stoichiometric amount of  $CpCo(CO)_2$  to afford complexed cyclopentadienones regioselectively<sup>96</sup> (eqn 25). Metallocyclopentadienes<sup>10a</sup> and metallocyclobutenones97 have been used in the formation of cyclopentadienones from alkynes and carbonylmetal compounds regioselectively.



SCHEME 37.

Sesquiterpene illudol<sup>98</sup> 108 was obtained from 107 via intramolecular [2+2+2] cyclization of 106 (Scheme 37). It is interesting to note that [5.6.4] ring system is constructed during cyclization from an acyclic precursor.

Vollhardt and co-workers have shown that intramolecular cyclization of enedyene 109 (prepared by Nicholas reaction), containing a tetrasubstituted double bond, provides a diastereomeric mixture of spirocyclic diene 110a and 110b which can be converted via routine functional group manipulation to the antimicrobial diterpene stemodine<sup>99</sup> 111 (Scheme 38).



SCHEME 38.

Vollhardt has demonstrated that the precursor 114a and 114b for daunomycinone<sup>100</sup> can be synthesised by reacting diyne 112 with alkene 113 in the presence of  $CpCo(CO)_2$  (Scheme 39).



Daunomycinone

SCHEME 39.

Interestingly, enamide **115** and bis(trimethylsilyl) acetylene (BTMSA) were cocyclized in the presence of CpCo(CO)<sub>2</sub> to form diastereomeric complexes **116a** and **116b**, providing galanthan ring systems. These intermediates were transformed to  $\gamma$ -lycorane<sup>101</sup> **117** by routine synthetic operations (Scheme 40).



SCHEME 40.

Further exploration of this reaction by Vollhardt and co-workers has shown that one aromatic double bond of many heterocyclic ring systems is capable of incorporation into cyclohexadiene ring. Reaction with N-substituted heterocycles, imidazole, pyrrole, indole and uracil derivatives has shown that aromatic double bond can function as the alkene component in the cyclization. A [2+2+2] cycloaddition of pyrrole<sup>102</sup> **118** was carried out to afford fused dihydro indole **119** (can 26).



N-substituted imidazole<sup>103</sup> **120** was reacted with BTMSA to give cycloadduct **121** in high yields (Scheme 41).



SCHEME 41.

N-substituted indole 122 also reacts with BTMSA to provide CpCo-complex. This reaction provides an entry to 4a, 9a-dihydro 9H carbazole<sup>104</sup> 123 (eqn 27).



Indole derivative 124 on cocyclization with 125 gives cobalt complex 126 which on treatment with MnO<sub>2</sub> furnishes propellane  $^{105}$  127, and the latter rearranges to spirofused compound 128 on oxidative removal of the metal (Scheme 42).

Substituted uranil<sup>106</sup> **129** undergoes cycloaddition to give Co-complexes **130** which can lead to various nucleoside derivatives (eqn 28).

A novel synthesis of fused 2H-pyrans<sup>107</sup> **132** has been achieved via  $\eta^5$ -cyclopentadienylcobalt complex-induced [2+2+2] cycloadditions of the alkynes **131** with ketones both inter as well as intramolecularly (eqn 29).

Vollhardt and co-workers<sup>108</sup> have synthesised enantiomerically pure cyclopentadienyl cobalt complexes from chiral ligands obtained from naturally occurring terpenes and acids. The chiral cobalt complexes **133a-c** were efficient catalysts in providing high diastereomeric excess. They have used these complexes for the photolytic cyclication of unsymmetrical  $\alpha, \omega$ -diynes **134** to metal-complexed cyclopentadienones **135** (eqn 30).





Later, these complexes were also exploited for diastereoselective encline cyclization to complexed cyclohexadienes. Prochiral  $\alpha$ , $\delta$ , $\omega$ -enediynes 136 cyclize to chiral tricyclidiene complex 137 in the presence of 133a as diastereometric complexes in 58:42 ratio (eqn 31).





# 2. Nicholas reaction

A phenomenon of long-standing interest in organometallic chemistry is the tremendously enhanced stability of carbonium ion flanked by organometallic metal moieties<sup>109</sup>. While considerable attention has been focussed on the various possible modes of these stabilizations, the potential applications of these cations in organic synthesis have largely remained an uncharted area. The use of dicobalt octacarbonyl for the protection of a triple bond is well known and the realisation that triple bond-coordinated Co<sub>2</sub>(CO)<sub>6</sub><sup>+</sup> moicty dramatically enhances stability of propargylic carbocations<sup>110–112</sup> has led to the growth of the synthetic transformations now known as Nicholas reaction (Scheme 43). This methodology has found numerous applications in organic synthesis and some of the salient features of its utility are discussed.





SCHEME 43.

# 2.1. Reaction with nucleophiles

The enhanced stability of (propargyl)  $Co_2(CO)e^+$  cations has made them an attractive intermediate for a new bond formation on reaction with a wide range of nucleophiles. In all cases, attack by nucleophile occurs exclusively at the propargylic carbon, resulting in a versatile propargylation method subsequent to mild oxidative demetallation.

# 2.2. Aromatics

Electron-rich aromatic compounds<sup>113</sup> including anisole, phenol, N, N-dimethylaniline, etc., react at room temperature or even below with the (propargyl)  $Co_2(CO)_6^+$  complexes to afford C-propargylated aromatic compounds (Scheme 44).



SCHEME 44.

# 2 3. Reaction with X-dicarbonyls

Propargylated cobalt complexes as salts with HBF<sub>4</sub> or TiCl<sub>4</sub> react easily with  $\beta$ -diketones and  $\beta$ -ketoesters affording mono C-propargylated products in good yields<sup>114</sup> (Scheme 45). This selective reaction reflects the reversibility of coupling reaction and the steric bulk of the (propargyl) Co<sub>2</sub>(CO)<sub>6</sub><sup>+</sup> group<sup>115,116</sup>. Reactions of chiral cobalt complexes with the prochiral  $\beta$ -diketones were found to proceed with diastereoselectivities of 2:1 to 15:1.



SCHEME 45.

# 2.4. Reaction with allyl and enol silanes

Propargyl dicobalt hexacarbonyl cations couple with allylsilanes to give complexes of 1,5enyncs in satisfactory yields<sup>117,118</sup> (Scheme 46). In the presence of BF<sub>3</sub> etherate a cobaltcomplexed propargylic ether can undergo an intramolecular alkylation with an allylic silane to provide six-, seven- and eight-membered complexed cycloalkynes<sup>45</sup> (eqn 32).

SCHEME 46.



Schreiber and co-workers<sup>119</sup> have performed an exocyclic intramolecular alkylation of allylic silane to afford six-membered ring with complete stereocontrol. Oxidative decomplexation of extra-annular cobalt complex provided the acetylene **138** (eqn 33). The intermolecular version of the reaction provides high levels of diastereoselection for *syn*-alkylated products provided certain stereocontrol elements are maintained. The intramolecular alkylation reaction with allylic silanes affords either intra or extra-annular cobalt alkyne complexes<sup>45,110</sup>.



Caple and Smit<sup>120</sup> have reported the trapping of the cation formed by electrophilic addition to 1,3-enyne complexes with trimethylsilyl enol ethers or allylsilanes (eqn 34).



The alkylation of silyl enol ether<sup>45</sup> with the cobalt complex of propargylic methyl ether affords alkylated ketone. Cobalt complex can be removed from the products using trimethylamine-N-oxide or ferric nitrate while the stereochemical nature of the product is being preserved (eqn 35).



Cobalt-mediated cyclopentanone annulation was used as a new methodology to give the guaiane sesquiterpene skeleton 142 as in cyclocolorenone<sup>121</sup> which was synthesised from the intermediate 141 derived by the reaction of enol silane 140 with cobalt complex 139. Magnus and co-workers have carried out<sup>122</sup> an intramolecular Nicholas reaction on cobalt-complexed propargylic methyl ether to provide an access to calicheamicinone model systems (Scheme 47).

In the same way,  $\alpha$ -alkoxy cations derived from acetylenic acetals<sup>123</sup> have been found to combine effectively with the enol derivatives, allyl silanes and enol silanes in the presence of BF<sub>3</sub>Et<sub>2</sub>O to afford the  $\beta$ -alkoxyacetylenic ketone derivatives in excellent yield and modest to excellent *syn* stereoselectivity (eqn 36). The uncomplexed acetals do not undergo reaction at-78°C but between – 20 and 0°C reaction did occur affording the corresponding acetylenic ketone as a 1:1 mixture of *syn* and *anti* diastereomers. It is clear therefore that the metal fragment not only facilitates coupling but also has categorical effect on diastereoselectivity.



Although highly diastereoselective products have been achieved from complexed aldehyde or acetal precursors with enol and allyl nucleophiles<sup>124</sup> as mentioned above, facile racemization of these cations has previously thwarted attempts to develop general, enantioselective route to the diastereomers. In order to get enantioselective propargyltation, Nicholas and co-workers<sup>125</sup> have used enantiomerically pure propargylic alcohol and converted them to diastereomeric dicobalt propargylium Co<sub>2</sub>(CO)<sub>5</sub>L complexes, where L may be PPh<sub>3</sub> or P[OCH(CF<sub>3</sub>)<sub>2</sub>]<sub>3</sub>. After demetallation of the resulting alkylated complex, enantiomerically pure compounds, were obtained (Scheme 48).

# 2.5. Reaction with amines

The Nicholas reaction with carbon nucleophiles<sup>111</sup> has been explored to a great extent and apart from this, the oxygen-centered nucleophiles<sup>126</sup> were also used frequently. However, only little is known about Nicholas reaction with nitrogen nucleophiles. The earliest example, an unoptimized reaction of propargylic cobalt salt of HBF<sub>4</sub> with acetonitrile in the presence of sulfuric acid, dates from 1981<sup>127</sup> (eqn 37).



(a) Co<sub>2</sub>(CO)<sub>8</sub>; (b) P[OCH(CF<sub>3</sub>)<sub>2</sub>]; (c) CO; (d) (NH<sub>4</sub>)<sub>2</sub>Ce(NO<sub>3</sub>)<sub>6</sub>

SCHEME 48.

$$\begin{bmatrix} HC \equiv CCH_2OH \\ \vdots \\ Co_2(CO)_6 \end{bmatrix}^+ + BF_4 \xrightarrow{H_2SO_4} HC \equiv CCH_2NHCOMe \\ \vdots \\ Ch_3CN \xrightarrow{I} Co_2(CO)_6 \end{bmatrix}$$
(37)

In 1990, Japanese group<sup>128</sup> reported the N-propargylation of indole and a few other heterocycles. Indole reacted with propargylic alcohol- $Co_2(CO)_6$  complexes to give 3-(1,1-dimethylpropargyl) indole **143** (Scheme 49) whereas N-methoxycarbonyl-tryptamine **145** gave the corresponding N- and C-substituted derivative **146a,b** with (propargyl acetate)  $Co_2(CO)_6$  complex **144** (Scheme 50).



SCHEME 49.



The reactions of the [(HC=CCH<sub>2</sub>)Co<sub>2</sub>(CO)<sub>6</sub>] BF<sub>4</sub> have been carried out with a wide range of amines to give the corresponding propargylic amines<sup>129, 130</sup> (Scheme 51). The primary amines were simultaneously C-alkylated by protecting the–NH<sub>2</sub> group.



SCHEME 51.

Recently, tertiary amines<sup>131</sup> have been synthesised by a selective reaction from cobaltcomplexed propargyl cation using primary and secondary amines as nucleophiles (eqn 38).



# 2.6. Reaction with other organometallic nucleophiles

The reaction of several methyl-metallic compounds (CH<sub>3</sub>)<sub>3</sub> ML<sub>a</sub> with the propargylated cobalt complexes had been tried to produce the methylated derivatives of propargyl group containing compounds<sup>132,133</sup> (eqn 39). The most efficient method of coupling of propargyl cations with acetylenic group was via the reaction (alkyne)<sub>3</sub>Al with complexed propargyl acetates to form 1,4-diyne complexes<sup>134</sup> (Scheme 52).

$$\begin{bmatrix} HC \equiv C - CH_2 \\ Co_2(CO)_6 \end{bmatrix}^+ \xrightarrow[-78°C]{(CH_3)_3Al} HC \equiv C - CH_2 CH_3 \\ HC \equiv C - CH_2 CH_3 \\ -78°C Co_2(CO)_6 \\ 91\%$$
(39)

$$HC \equiv CCH_{2} - OAC + (n - BuC \equiv C)_{3} AI \xrightarrow{CH_{2}CI_{2}}_{-78^{\circ} \rightarrow 0^{\circ}C} HC \equiv CCH_{2} - C \equiv C - n - Bu$$

$$HC \equiv CCH_{2} - C \equiv C - n - Bu$$

$$HC \equiv CCH_{2} - C \equiv C - n - Bu$$

$$HC \equiv CCH_{2} - C \equiv C - n - Bu$$

$$HC \equiv CCH_{2} - C \equiv C - n - Bu$$

SCHEME 52.

## 2.7 Miscellaneous nucleophiles

Although not much work has been done on the reaction of propargyl  $Co_2(CO)_6^+$  cation with non-carbon-centered nucleophiles, Siegel *et al*<sup>134,135</sup> found that secondary alkyl acetylenes car be prepared through the reduction of the corresponding cobalt-complexed  $\alpha$ -acetylenic alcohols with NaBH<sub>4</sub>/CF<sub>3</sub>COOH in dichloromethane (Scheme 53). Deuterium-labelled disopropylacetylene prepared by this method has been used in the synthesis of (hexaisopropyld<sub>6</sub>) benzene.



# 2.8. $\alpha$ -Alkoxy cations

The highly reactive (1,2-epoxy alkyne) dicobalt hexacarbonyl complexes **147** had been generated *in situ* by treatment of 1,2 epoxy-3-alkyne (from 1-octene-3-yne/MCPBA) and with a slight deficiency of  $Co_2(CO)_8$  in benzene solution at 5°C. The epoxide reacts with several nucleophiles (*i.e.*, CH<sub>3</sub>OH, H<sub>2</sub>O, Cl<sub>3</sub>COOH) under acidic conditions to produce the 1-substituted 2-hydroxy products<sup>136,137</sup> in good yields (Scheme 54).



SCHEME 54.

## COBALT CARBONYLS

# 2.9. a-Vinyl cations

In order to further elucidate the steric and eletronic properties of the propargylic carbonium ion stabilized with Co<sub>2</sub>(CO)<sub>6</sub>, the reactions of various nucleophiles with the vinylogous cations<sup>138</sup> were examined. Accordingly, it was demonstrated that anisole<sup>139</sup> reacts regio- and stereoselectively with **148** to give (E)-1,3-enyne **149** derivative in good yields (eqn 40).



# 2.10. a-Cyclopropyl cations

Descoins and Samain<sup>140</sup> have shown the contrast between the stereoselectivitics of the reaction with the free and complexed cyclopropyl carbinols. It shows that attachment of complex not only facilitates the reaction but also provides a highly stereoselective (E)-1,3-enyne formation. Saha<sup>141</sup> has extended this reaction to carbon nucleophiles (*i.e.*, allyl silanes, anisole, vinylacetate, etc.) which reacted without cleavage of cyclopropane ring (Scheme 55).



SCHEME 55.

## JAVED IQBAL et al

## 3. Carbonylation reaction

Carbonylation<sup>1,13</sup>, as the name suggests here, involves the process of introducing CO into the molecule. Transition metal-promoted carbonylation<sup>142,143</sup> of olefins, acetylenes, halides, alcohols, amides, nitro compounds, etc., are very important in both industrial and academic research. Cobalt carbonyls have been widely used and most extensively studied among the metal carbonyls. It catalyses the hydroformylation of olefin, and was first discovered by Roslen in 1937 (eqn 41). Olefins react with  $Co_2(CO)_8$  in the presence of  $CO/H_2$  to provide aldehydes<sup>144</sup> in very high yields. These reactions were later on developed into an useful industrial process<sup>145</sup> for the synthesis of aldehydes from alkenes (Scheme 56).

$$>c=c < \frac{Co_2(CO)_8}{CO + H_2}$$
  $H > C - C < CHO$  (41)



SCHEME 56.

The carbonylation process using  $Co_2(CO)_8$  is more useful with vinyl ethers<sup>146</sup> or vinyl acetates which leads to the formation of mainly one regioisomer (eqn 42). These reactions have found widespread application on carbohydrate substrates owing to their high regioselectivity and mild conditions (eqn 43).





Heck and co-workers have developed a novel route to alkyl-Co(CO)<sub>4</sub> complex<sup>147</sup> from the reaction between alkyl halide or sulphonates and Co<sub>2</sub>(CO)8. The alkyl cobalt complex thus prepared underwent CO insertion to give acyl cobalt complex which was converted to aldehydes, amides or esters on reaction with hydrogen, amine or alcohols, respectively (Scheme 57).



SCHEME 57.

Conjugated dienes undergo reductive hydroformylation to yield saturated monoaldehyde148 whereas non-conjugated dienes are prone to form ketones as byproduct (eqn 44).

CH<sub>2</sub>=CH-CH=CH<sub>2</sub>+ CO + H<sub>2</sub> 
$$\xrightarrow{CO_2(CU)_8}$$
 CH<sub>3</sub>(CH<sub>2</sub>)<sub>3</sub>CHO + CH<sub>3</sub>CH<sub>2</sub>CH(CH<sub>3</sub>)CHO (44)

Secondary and tertiary alcohols readily undergo hydroformylation presumably via the corresponding olefins (eqn 45).

PhCH<sub>2</sub>OH + H<sub>2</sub> + CO 
$$\xrightarrow{Co_2(CO)_8}$$
 PhCH<sub>2</sub>CH<sub>2</sub>OH + PhCH<sub>3</sub>  
31% 63% (45)

Dilactone<sup>149</sup> can be synthesised by Co-catalysed carbonylation of acetylenes via the isolable complexes  $Co_2(CO)_6C_2H_2$  and  $Co_2(CO)_9C_2H_2$  as intermediates (eqn 46).

$$HC \cong CH + CO \xrightarrow{Co_2(CO)_8} O \xrightarrow{O} O \xrightarrow{O} O \xrightarrow{(46)} 70\%$$

Alper and co-workers<sup>150</sup> have synthesised  $\gamma$ -hydroxy lactone using acyl-Co-complexes derived from the reaction of Co<sub>2</sub>(CO)<sub>8</sub> with CH<sub>3</sub>I and CO. Co-complex reacts with alkyne to give 4-keto-3-alkenoyl cobalt intermediate, using phase transfer catalyst. This complex gives unsaturated keto acid that cyclised to give  $\gamma$ -hydroxy lactone (eqn 47).

$$PhC \equiv CH \xrightarrow{cat.Co_2(CO)_8, Mel}_{5N NaOH, CTAB, C_6H_6} Me \xrightarrow{OH}_{OH} OH (47)$$

One of the outstanding developments in the area of carbonylation using  $Co_2(CO)_8$  is the contribution from Murai and co-workers<sup>[51]</sup>. They have developed a direct method for the synthesis of enolsilyl ether from cyclic olefins in the presence of CO and diethyl (methyl) silane (eqn 48). They have suggested a catalytic pathway for siloxymethylation.

$$(48)$$
 + CO + HSiEt<sub>2</sub>Me  $Co_2(CO)_8$  OSiEt<sub>2</sub>Me  $(48)$ 

High affinity of silicon towards oxygen in the key intermediate  $R_3SiC_0(CO)_4^{152}$  is the driving force for the cleavage of C–O bond in oxygenated compound **148** to give intermediate **149** having carbon-cobalt bond. Insertion of CO gives acyl cobalt complex **150** which reacts with HSiR<sub>3</sub> to give Co-complex **151** and the latter on reductive elimination of  $R_3SiC_0(CO)_3$  provides aldehyde. Subsequently the reductive addition of HSiR<sub>3</sub> gives Co-complex **152**, which on elimination of HSiC<sub>0</sub>(CO)<sub>3</sub> provides enoisilane **153** (Scheme **58**).

Reaction of epoxides with Co<sub>2</sub>(CO)<sub>8</sub>/CO depends upon the solvent used in the reaction as 3-hydroxy esters<sup>146</sup> were obtained by using MeOH as solvent whereas in aprotic solvent  $\alpha$ , $\beta$ -unsaturated acids were found to be the major products (Scheme 59). This methodology has been used for the synthesis of 2-(6-methoxycarbonylhexyl)-cyclopent-2-ene-1-one **154**. This is a simple and short route for the synthesis of **154** via Co-catalysed carbonylation of intermediate epoxide. The reaction does not occur in the absence of base and ethanol was used for achieving the highest selectivity (Scheme 60). When reaction of epoxides was (Scheme 61).



Scheme 58.



Scheme 59.

JAVED IQBAL et.al



SCHEME 60

Later, Murai and co-workers have reported hydroformylation of cyclic ethers<sup>154</sup> in the presence of hydrosilanes. It has been shown that direct hydroformylation of these molecules suffers from undesirable side reactions.



SCHEME 61.

Tetrahydrofuran, oxetane and 1,2-epoxycyclohexane undergo cleavage with diethyl (methyl) silane and CO to give silyl-protected hydroxy aldehydes<sup>155</sup> (Scheme 62). In the absence of CO, epoxides are rearranged to ketone<sup>1</sup> by  $Co_2(CO)_8$  (Scheme 63).



452



SCHEME 63.

Oxetanes react with  $Co_2(CO)_8$  and CO to give 4-hydroxy acyl cobalt tetracarbonyls which decompose to give  $\gamma$ -lactone<sup>156</sup>. Large ring lactones can also be prepared by using chloroalcohol (Scheme 64). Allyl alcohols on intramolecular cyclization give lactones (eqn 49).



SCHEME 64.

$$\longrightarrow OH \qquad \xrightarrow{\text{Co}_2(\text{CO})_{\text{B}}/\text{CO}} \qquad \xrightarrow{\text{O}} \qquad (49)$$

Alkyl and acyl Co-complexes react with 1,3-dienes to provide  $\eta^3$ -allyl derivatives<sup>150</sup> which decompose to give 1-acyl, 1,3-dienes (eqn 50).



1,2-Bis(siloxy) olefin<sup>157</sup> can be prepared from Co-catalysed reaction of aldehydes with HSiR<sub>3</sub> in the presence of PPh<sub>3</sub>. The PPh<sub>3</sub> as co-catalyst is necessary to avoid undesired hydrosilylation of aldehydes (eqn 51). In these reactions, 3-fold excess of HSiEt<sub>2</sub>Me was used.

RCHO + HSiEt<sub>2</sub>Me + CO 
$$\xrightarrow{\text{cat. Co}_2(\text{CO})_8}$$
  $\xrightarrow{\text{MeEt}_2\text{SiO}}$   $\xrightarrow{\text{OSiEt}_2\text{Me}}$   $\xrightarrow{\text{Cat. PPh}_3}$   $\xrightarrow{\text{R}}$   $\xrightarrow{\text{R}}$   $\xrightarrow{\text{H}}$   $\xrightarrow{\text{H}}$   $\xrightarrow{\text{(51)}}$   $\xrightarrow{\text{R}}$ 

Later, these authors reported conversion of aldehydes to their higher  $\alpha$ -siloxy aldehydes<sup>158</sup> by hydrosilane and CO. Here the use of an excess of starting aldehyde is essential to avoid formation of 1.2-bis(siloxy) alkenes (eqn 52).



These reactions may be proceeding via silyl cobalt complex 155 formed in situ from  $Co_2(CO)_8$  with hydrosilane. The intermediate is  $\alpha$ -siloxy alkyl cobalt compound 156 formed in situ from 155 with aldehyde. The high affinity of silyl group for oxygen may force C–Co bond formation (eqn 53). Murai and co-workers<sup>159</sup> have described transformation of alkyl acctates to [(trialkylsiloxy)methylene]alkanes (Scheme 65).



This methodology is also applicable to lactone which is converted to the corresponding silyl enol ethers obtained by reductive opening of the ring (Scheme 66).

Later, these workers have reported a cobalt carbonyl-catalysed ring enlargement of cyclobutanones<sup>160</sup> with hydrosilanes and CO. This was the first example reported for the catalytic incorporation of CO into a ketonic carbon (Scheme 67). This reaction provides a



Scheme 66.

novel method for the formation of five-membered rings containing disiloxy alkene which can be useful in the synthesis of polycyclopentanoids.



**SCHEME 67.** 

The authors have also reported that  $[R_3SiCo(CO)_4]$  is efficient catalyst for nucleophilic oxymethylation<sup>161</sup> of oxiranes to give 1,3-diol derivatives (Scheme 68). It was observed that functional groups present in oxiranes are not affected under these reaction conditions.



SCHEME 68.

Murai and co-workers have reported a novel route for the synthesis of C-glycosyl compounds from glycosyl acetates via glyoxymethylation<sup>162</sup> (Scheme 69). C-Glycosyl compounds are valuable as multipurpose building blocks and also as intermediate for methylene phosphonate and homo-C-nucleosides. This method is useful for one carbon chain extension at the anomeric centre of glycosides.



SCHEME 69

Ito<sup>163</sup> has utilised this methodology for the synthesis of 2-deoxy-C-nucleoside skeletons (Scheme 70). It was noted that siloxymethyl group has been introduced *trans* to the adjacent (C-3) acetate group which is in consonance with Murai's result.



SCHEME 70.

Foa and co-workers<sup>164</sup> have shown that in the presence of appropriate base alkyl tetracarbonyl cobalt complexes catalyse the carboxylation of aryl halides in aliphatic alcohols to provide ester. In contrast, Murai and co-workers have shown that the reaction of aryl halides under phase transfer conditions in the presence of MeI and NaOH have a mixture of aryl methyl ketones and aromatic carboxylic acid (eqn 54). This reaction proceeds *via* methyl tetra carbonyl cobalt complex<sup>165</sup> which can be generated *in situ* from Co<sub>2</sub>(CO)<sub>8</sub> with MeI. Product composition is highly dependent on base and solvent used.

$$\operatorname{Arx} \xrightarrow{\operatorname{Co}_2(\operatorname{CO})_8/\operatorname{CO}/\operatorname{Me}/\operatorname{NaOH}}_{\operatorname{C}_6\operatorname{H}_6\operatorname{-H}_2\operatorname{O}/\operatorname{CTAB}} \operatorname{ArCOMe} + \operatorname{ArCO}_2\operatorname{H}$$
(54)

Later, Miura and co-workers<sup>166</sup> have demonstrated the carbonylation of vinyl halides on the corresponding carboxylic acid under these conditions (eqn 55).



This methodology can be employed for the synthesis of Furan-2(5H)-ones<sup>167</sup> by carbonylation of 3-chloroprop 2-enols. When carbonylation was carried out in the presence of benzaldehyde using NaOH, an adduct was formed in 70% and furanone in 13% yield. These products were further converted into  $\gamma$ -alkyldenebutenolides on treatment with thionyl chloride in pyridine (Scheme 71).





Alper and co-workers<sup>168</sup> have developed an efficient method for regiospecific acylation of fulvenes by using phase transfer agent (eqn 56). Later, these workers<sup>169</sup> have shown the hydroxyacylation of allenes under similar conditions (eqn 57).



Cobalt carbonyl is an efficient catalyst for carbonyl insertion reactions between C-N and N-N double or triple bond. Unsaturated amines and amides give lactams and imides<sup>170</sup>, respectively, under the aegis of cobalt carbonyl and CO (Scheme 72). Schiff bases and azo compounds<sup>171</sup> provide phthalimidines and 2-phenyl indazolone, respectively, by cyclocarbonylation reaction with Co<sub>2</sub>(CO)<sub>8</sub>/CO (Scheme 73). Phenyl hydrazones and oximes also undergo cyclocarbonylation reactions to give cyclic amides (Scheme 74).



458

Aldehydes and amides in the presence of  $Co_2(CO)_8/CO$  provide N-acyl aminoacids<sup>172</sup> (eqn 58). Trifluorovaline and trifluoronovaline are synthesised *via* cobalt-catalysed amidocarbonylation<sup>142</sup> of 2-TFMPA and 3-TFMPA, respectively, which are further hydrolysed to give free amino acid (Scheme 75).



Ketones can be prepared by the reaction of organomercury compounds  $^{173}$  in the presence of Co<sub>2</sub>(CO)<sub>8</sub>/CO (eqn 59).



# 4. Miscellaneous reactions

Murai and co-workers<sup>174</sup> have developed a novel and efficient method for the synthesis of N, N-disilylamines by reduction of aromatic nitriles using cobalt carbonyl-catalysed addition of two molecules of HSiMe<sub>3</sub>.

Aliphatic nitriles did not react with HSiMe<sub>3</sub> whereas in p-(cyanomethyl)-benzonitriles, the cyano group adjacent to benzene ring, selectively reacts with HSiMe<sub>3</sub>. The rate of conversion of aromatic nitriles having electron-withdrawing group or sterically hindered nitriles is rather low (Scheme 76).





SCHEME 76.

Chatani and co-workers<sup>175</sup> have developed a method for the formation of pyrrole ring from alkynes and cyanotrimethyl silane in the presence of  $CO_2(CO)_8$  (Scheme 77).



SCHEME 77.

Isobe and co-workers<sup>176</sup> have recently shown the epimerization of C-1 alkynyl group on pyranose ring through cobalt complexes under acidic conditions. Thus, **157** on complexation with  $Co_2(Co)_8$  provides Co-complex **158** which was equilibrated under acidic conditions by using TfOH to give the opposite isomer in very high yields. This reaction was carried out under various conditions and best results were obtained at higher temperatures with catalytic amount of iodine which afforded the isomer in very high yields. A similar transformation is also achieved on pyranose ring **159** containing substituent with two triple bonds by epimerisation of bis-cobalt complex **160** (Scheme 78).

# 5. Conclusion

The foregoing sections have clearly established the versatility of cobalt carbonyl in contemporary synthesis and this development has a very strong bearing on the future attempts towards pursuit of selectivity during the construction of sensitive and complex organic structures. Pauson–Khand, Nicholas and Vollhardt reactions are the outstanding features of these endeavours which will go a long way in achieving the desired efficiency and selectivity which has now become a hallmark of modern synthesis.





SCHEME 78.

# Acknowledgement

We are grateful to Mr T. Punniyamurthy for his help during the preparation of this manuscript.

# References

1.	Davies, S. G.	Organotransition metal chemistry Applications to organic synthesis, 1986, Pergamon Press.
2.	WULFF, W. D.	In Comprehensive organic synthesis (Trost, B. M. and Fleming, I., eds), 1991. Vol. 5, p.1065, Pergamon Press.

# JAVED IQBAL et al

- PARSHALL, G.W. 3.
- 4. EVANS, D., YAGUPSKY, G. AND WILKINSON, G
- 5 MILLER, S. A.
- INDICTOR, N. AND BRILL, W.F. 6
- 7 DOTZ, K.H., FISCHER, H., HOFMAN, P , KREISSEL, F.R., SCHUBERT, U. AND WEISS, K.
- 8. a. WANG, S.L.B., SU, J., WULFF, W.D. AND HOOGSTEEN, K.
  - b. WULFF, W.D. AND XU, Y.C.
  - C. WULFF, W.D., YANG, D.C. AND MURRAY, C.K.
  - d. WULFF, W.D. AND TANG, P.-C.
  - e. BAO, J., DRAGISICH, V., WENGLOWSKY, D. AND WULFF, W.D.
  - f. CHAMBERLIN, S. AND WULFF, W. D.
  - g. ANDERSON, B A., WULFF, W.D. POWERS, T.S., TRIBBITT, S. AND RHEINGOLD, A.L.
- 9. a. DOTZ, K.H. AND POPALL, M.
  - b. Dotz, K.H.
  - C. DOTZ, K.H. AND KUHN, W.
  - d. DOTZ, K.H., WERNER, K., MUELLER, G., Angew. Chem Int. Edn Engl., 1986, 25, 812-817. HUBER, B. AND ALT, H.G.
  - e. Dotz, K.H., NOACK, R. AND MULLER, G.
  - f. DOTZ, K.H., NOACK, R., HARMS, K. AND MUELLER, G.
  - g. GROTJAHN, D.G. AND DOTZ, K.H.
  - h. DOTZ, K.H., GROTJAHN, D. AND HARMS, K
- 10. a. COLMAN, J.P. AND HEGEDUS, L.S.
  - b. SCHWINDT, M.A., LEJON, T. AND HEGEDUS, L.S.
  - C. IMWINKELREID, R. AND HEGEDUS, L.S.
  - d. HEGEDUS, L.S., MONTGOMERY, J., NARUKAWA, Y. AND SNUSTAD, D.C.
  - e. HEGEDUS, L.S., SCHWINDT, M.A., DE LOMBAERT, S. AND IMWINKELRIED, R.
  - f. HEGEDUS, L. S. AND SODERBERG, B.C.
  - g. HEGEDUS, L.S., BATES, R.W. AND SODERBERG, B.C.

- J. Mol Catal., 1978, 4, 243.
- J. Chem Soc A, 1968, 2660-2665.
- Chem. Process Engng, 1969, 50, 63.
- J Org Chem, 1965, 30, 2074-2075.

Transition metal carbene complexes, 1984, Verlag Chemie, Deerfield Beach, FL,

- J Am. Chem. Soc., 1992, 114, 10665-10666.
- J. Am. Chem. Soc., 1988, 110, 2312-2314. J. Am. Chem. Soc., 1988, 110, 2653-2655.
- J. Am. Chem Soc., 1984, 106, 434-436,
- J. Am. Chem. Soc., 1991, 113, 9873-9875.
- J. Am. Chem. Soc., 1992, 114, 10667-10669. J. Am. Chem. Soc., 1992, 114, 10784-10978.
- Tetrahedron, 1985, 41, 5797--5802.
- Angew. Chem. Int Edn Engl., 1984, 23, 587-608.
- J. Organomet Chem., 1985, 286, C23.
- J Chem. Soc., Chem. Commun, 1988, 302-304.

Tetrahedron, 1990, 46, 1235-1252.

Synlett, 1991, 381. Angew. Chem. Int Edn Engl., 1989, 28, 1384.

Principles and applications of organotransition metal chemistry 1980, University Science Books, Menlo Park.

Organometallics, 1990, 9, 2814.

- Organometallics, 1988, 7, 702. J. Am Chem Soc., 1991, 113, 5784-5791.
- J. Am. Chem. Soc., 1990,112, 2264-2273.

J. Org. Chem., 1991, 56, 2209-2212.

J. Am. Chem. Soc., 1991, 113, 923-927.



## COBALT CARBONYLS

11.		MERLIC, C.A. AND XU, D.	J Am Chem. Soc., 1991, 113, 9855-9856
12.		HARVEY, D F. AND LUND, K.P.	J Am Chem Soc, 1991, 113, 5066-5068.
13.		WENDER, I. AND PINO, P.	Organic synthesis via metal carbonyls, Vol. 1, 1968, Vol. 2, 1977, Wiley.
14.		KHAND, I.U., KNOX, G.R., PAUSON, P.L., WATTS, W.E. AND FOREMAN, M.I	J Chem Soc, Perkin Tians 1, 1973, 977-981
15.		Khand, I.U., KNOX, G.R., Pauson, P.L. and Watts, W.E.	J Chem Soc., Chem. Commun, 1971, 36.
16		PAUSON, P.L. AND KHAND, I.U.	Ann. NY Acad Sci., 1977, 295, 2.
17.		JAFFER, H.J. AND PAUSON, P.L.	J. Chem Res. (M), 1983, 2201.
18		Bladon, P , Khand, I.U. and Pauson, P.L.	J Chem. Res. (M), 1977, 153
19.		Schore, N.E.	Chem Rev, 1988, 88, 1081-1119.
20.		Schore, N.E.	Org React, 1991, 40, 1.
21.		KHAND, I.U. AND PAUSON, P.L.	J. Chem. Res (M), 1977, 168.
22.		Khand I U., Mahaffy, C.A.L. and Pauson, P.L.	J. Chem. Res (M), 1978, 4454.
23.		KHAND, I.U., MURPHY, E AND PAUSON, P.L.	J Chem Res (M), 1978, 4434.
24.		Khand, I.U., Pauson, P.L. and Habib, M.J.A.	J Chem Res (M), 1978, 4418.
25.		Krafft, M.E.	J Am Chem Soc, 1988, 110, 968-970.
26.		Schore, N.E.	Synth. Commun., 1979, 9, 41-47.
27.		La Belle, B.E., Knudsen, M.J., Olmstead, M.M., Hope, H., Yanuck, M.D. and Schore, N.E.	J. Org Chem., 1985, 50, 5215-5222.
28.		SAMPATH, V., LUND, E.C., KNUDSEN, M. J., OLMSTEAD, M.M. AND SCHORE, N.E.	J Org. Chem., 1987, 52, 3595-3603.
29	a.	Keyaniyan, S., Apel., M., Richmond, J.P. and de Meijere, A.	Angew. Chem. Int. Edn Engl., 1985, 24, 770-771.
	b.	Liese, T. and de Meijere, A.	Chem. Ber., 1986, 119, 2995-3026.
	c.	DE MEIJERE, A.	Chem. Ber., 1987, 120, 865.
30.		Montana, A. M., Moyano, A., Pericas, M.A. and Serratosa, F.	Tetrahedron, 1985, 41, 5995-6003.
31.		Smit, W.A., Kirbev, S.L., Nefedov , O.M. and Tarasov, V.A.	Tetrahedron Lett, 1989, 30, 4021-4024.
32.		Krafft, M.E., Juliano, C.A., Scott, I.L., Wright, C. and McEachin, M.D.	J. Am. Chem Soc., 1991, 113, 1693-1703.
33.		JEONG, N., YOO, SE., LEE, S.J., LEE, S.H. AND CHUNG, Y.K.	Tetrahedron Lett, 1991, 32, 2137-2140.
34.		SCHORE N.E. AND CROUDACE, M.C.	J. Org. Chem, 1981, 46, 5436-5438.

#### JAVED IQBAL et al

- 35. a. HUA, DH.
  - b. HUA, D H , COULTER, M.J. AND BADEJO, I.
- Seto, H., Sasaki, T., Yonchara, H. and Uzawa, J.
- MAGNUS, P., EXON, C. AND Albaugh-Robertson, P.
- 38. a. MAGNUS, P. AND QUAGLIATO, D.A.
  - b. EXON, C. AND MAGNUS, P.
  - C. MAGNUS, P. AND QUAGLIATO, D. A.
- 39. a. MAGNUS, P. AND PRINCIPE, L.M.
  - MAGNUS, P., PRINCIPE, L.M. AND SLATER, M.J.
- 40. MAGNUS, P. AND BECKER, D.P.
- 41. a. KNUDSEN, M.J. AND SCHORE, N.E
  - b. Schore, N.E. and Knudsen, M.J.
- 42. Schore, N.E. and Rowley, E.G.
- a. Carcellar, E., Centellas, V., Moyano, A., Pericas, M.A. and Serratosa, F.
  - b. Carceller, E., Garica, M.L., Moyano, A., Pericas, M.A. and Serratosa, F.
  - c. Almansa, C., Carceller, E., Gracia, E., Torrents, A. and Serratosa, F.
- 44. BILLINGTON, D.C. AND WILLISON, D.
- 45. Schreiber, S.L., Sammakia, T. and Crowe, W.E.
- Smit, W.A., Gybin, A.S., Shashkov, A.S., Strychkov, Y.T., Kyz'mina, L.G., Mikaelian, G.S., Caple, R. and Swanson, E.D.
- a. Smit, W.A., Gybin, A.S., Simonyan, S.O., Shashkov, A.S., Tarasov, V.A. and Ibragimov, I.I.
  - b. Smit, W.A., Simonyan, S.O., Shashkov, A.S., Mamyan, S.S., Tarasov, V.A. and Ibragimov, I.I.
- VERETENOV, A.L., SMIT, W.A., VORONTSOVA, L.G., KURELLA, M.G., CAPLE, R. AND GYBIN, A.S.
- 49. SHAMBAYATI, S., CROWE, W.E. AND SCHREIBER, S.L.

J. Am. Chem Soc., 1986, **108**, 3835–3837. Tetrahedron Lett., 1987, **28**, 5465–5468.

Tetrahedron Lett., 1978, 19, 923-926.

Tetrahedron, 1985, 41, 5861-5869.

- Organometallics, 1982, 1, 1243.
- J. Am. Chem. Soc , 1983, 105, 2477-2478.
- J Org. Chem., 1985, 50, 1621~1626.
- Tetrahedron Lett , 1985, **41**, 4851–4858. J. Org. Chem., 1987, **52**, 1483.
- J Am Chem. Soc., 1987, 109, 7495-7498.
- J. Org. Chem., 1984, 49, 5025-5026.
- J. Org. Chem., 1987, 52, 569-580.
- J. Am. Chem. Soc., 1988, 110, 5224-5225.
- Tetrahedron Lett., 1985, 26, 2475-2476.

Tetrahedron, 1986, 42, 1831-1839.

Synth. Commun., 1988, 18, 381-390.

- Tetrahedron Lett., 1984, 25, 4041-4044. J. Am. Chem. Soc., 1986, 108, 3128-3130.
- Tetrahedron Lett., 1986, 27, 1241.

Izv. Akad. Nauk SSSR, Ser. Khim., 1985, 2650.

Izv. Akad. Nauk SSSR, Ser. Khim., 1987, 234.

Tetrahedron Lett., 1991, 32, 2109-2112.

Tetrahedron Lett., 1990, 31, 5289-5292.

.

# COBALT CARBONYLS

50.		Jeong, N., Lee, B.Y., Lee S.M., Chung, Y.K. and Lee, S G.	Tetrahedron Leii , 1993, 34, 4023–4026.
51.		HOYE, T.R. AND SURIANO, J.A.	J. Org. Chem., 1993, 58, 1659-1660.
52.		SMIT, W.A., BUCKANJUK, S.M., SIMONYAN, S.O., SHASHKOV, A.S., CAPLE, R., GYBIN, A.S., ANDERSON, L.G. AND WHITEFORD, J.A.	Tetrahedron Lett , 1991, <b>32</b> , 2105–2108.
53.		Krafft, M.E., Scott, I.L. and Romero, R.H.	Tetrahedron Lett., 1992, 33, 3829–3832.
54.		Vollhardt, K.P.C.	Angew. Chem., Int. Edn Engl., 1984, 23, 539-556.
55.	a.	Aalbersberg, W.G.L., Barkovich, A.J., Funk, R.L., Hillard III, R.L. and Vollhardt, K.P.C.	J. Am. Chem. Soc., 1975, 97, 5600–5602.
	b.	HILLARD III. R.L. AND Vollhardt, K.P.C.	Angew. Chem. Int. Edn Engl., 1975, <b>14</b> , 712–713; Angew. Chem., 1975, <b>87</b> , 744–745.
	c.	HELARD III, R.L. AND VOLLHARDT, K.P.C.	J Am. Chem Soc., 1977, 99, 4058-4069.
56.		Vollhardt, K.P.C.	Acc. Chem. Res., 1977, 10, 1.
57.		Gesing, E.R.F., Sinclair, J.A. and Vollhardt.,K.P.C.	J. Chem Soc., Chem. Commun., 1980, 286-287.
58.	a.	Wakatsuki, Y., Nomura, O., Kitaura, K., Morokuma, K. and Yamazaki, H.	J. Am. Chem. Soc., 1983, 105, 1907–1912.
	b.	YAMAZAKI, H. AND WAKATSUKI, Y.	J. Organomet. Chem., 1977, 139, 157-167.
	c.	YASUFUKU, H. AND YAMAZAK', H.	J. Organomet. Chem., 1977, 127, 197-207.
59.		MCALISTER, D.R., BERCAW, J.E. AND BERGMAN, R.G.	J. Am. Chem. Soc., 1977, 99, 1666-1668.
60.		MCDONELL BUSHNELL, L.P., EVITT, E.R. AND BERGMAN, R.G.	J. Organomet. Chem., 1978, 157, 445-456.
61.		Chang, CA., Francisco, C.G., Gadek, T.R., King, J.A. Jr, Sternberg, E.D. and Vollhardt, K.P.C.	In Organic synthesis. Today and tomorrow, (Trost, B.M. and Hutchinson, C.R., eds), 1981, p.71, Pergamon Press.
62.		HALTERMAN, R.L., NGUYEN, N.H. AND VOLLHARDT, K.P.C.	J. Am. Chem. Soc., 1985, 107, 1379–1387.
63.	a.	EKHADEM, H.S.	Anthracycline antibiotics, 1982, Academic Press.
	ь.	Arcamone, F.	Med. Chem., 1981, 17, 31.
	c.	Kelly, T.R.	Ann. Rep. Med. Chem., 1979, 14, 288.
64.	a.	Shamma, M.	The isoquinoline alkaloids, 1972, Academic Press.
	b.	SHAMMA, M. AND MONIOT, J.L.	Isoquinoline alkaloid research 1972-1977, 1978, Plenum Press.
65.		HILLARD III, R.L., PARNELL, C.A. AND VOLLHARDT, K.P.C.	Tetrahedron, 1983, <b>39</b> , 905.
66.		Vollhardt, K.P.C.	Pure Appl. Chem., 1980, 52, 1645-1667.
67.		HILLARD III, R.L. AND VOLLHARDT, K.P.C.	J. Am. Chem. Soc., 1976, 98, 3579–3582.

- 68. FUNK, R.L. AND VOLLHARDT, K.P C.
- 69 PERKINS, P. AND VOLLHARDT, K P.C.
- 70 VOLLHARDT, K.P C.
- 71. a HATFIELD, W.E.
  - b. WEGNER, G.
  - c. BAUGHMAN, R.H., BREDAS, J.L., CHANCE, R.R., ELSENBAUMER, R.L AND SHACKLETTE, L.W.
- 72. FUNK, R.L. AND VOLLHARDT, K.P.C.
- 73. Reviews
  - a. QUINKERT, G. AND STARK, H.
  - b. FUNK, R.L. AND VOLLHARDT, K.P.C.
  - c. KAMETANI, T.
- a. WAKATSUKI, Y., KURAMITSU, T. AND YAMAZAKI, H.
  - b. WAKATSUKI, Y. AND YAMAZAKI, H.
- DUNACH, E., HALTERMAN, R.L. AND VOLLHARDT, K.P.C.
- GADEK, T.R. AND VOLLHARDT, K.P.C
- 77. MARTIN, S.F
- 78. MALACRIA, M. AND VOLLHARDT, K.P.C.
- CHANG, C.A., KING, J.A. AND VOLLHARDT, K.P.C.
- 80. a. FUNK, R.L. AND VOLLHARDT, K.P.C.

#### b. VOLLHARDT, K.P.C.

- Sternberg, E. D. and Vollhardt, K.P.C.
- 82. HIRAI, K. AND KISHIDA, Y.
- 83. a. Ananchenko, S.N. and Torgov, I.V.
- b. Zakharychev, A.V., Ananchenko, S.N. and Torgov, I.V.
- CLINET, J.-C., DUNACH, E. AND VOLLHARDT, K.P.C.
- RUFER, C., SCHRODER, E. AND GIBIAN, H.

Angew Chem, 1976, 88, 63; Angew Chem, Int Edn Engl., 1976, 15, 53.

Angew Chem, 1978, 90, 643; Angew Chem, Int. Edn Engl., 1978, 17, 615–616.

Top Curr. Chem , 1975, 59, 113.

Molecular metals, 1979, Plenum Press.

Angew Chem., 1981, 93, 352; Angew. Chem., Int Edn Engl., 1981, 20, 361-381.

Chem Rev., 1982, 82, 209-222.

J Chem. Soc., Chem Commun, 1976, 833-834; J. Am Chem. Soc., 1976, 98, 6755-6757 and 1980, 102, 5245-5253.

Angew. Chem., 1983, 95, 651; Angew Chem., Int. Edn Engl., 1983, 22, 637–655

- Chem. Soc. Rev., 1980, 9, 41.
- Pure Appl Chem., 1979, 51, 747.
- Tetrahedron Lett., 1974, 4549-4552.

J Organomet Chem., 1977, 139, 169-177.

J Am. Chem Soc., 1985, 107, 1664-1671.

Angew. Chem., 1981, **93**, 801; Angew. Chem., Int. Edn Engl, 1981, **20**, 802–804.

- Tetrahedron, 1980, 36, 419.
- J. Org Chem., 1984, 49, 5010-5012.
- J. Chem. Soc., Chem. Commun, 1981, 53-55.

J. Am. Chem. Soc., 1977, 99, 5483-5484; 1979, 101, 215-217; 1980, 102, 5253-5261.

Ann. N.Y. Acad. Sci., 1980, 333, 241.

J Org. Chem., 1982, 47, 3447-3450; 1984, 48, 1574-1583.

Tetrahedron Lett., 1972, 2117–2120. Tetrahedron Lett., 1963, 1553–1558. Steroids, 1964, **4**, 31.

J. Am. Chem. Soc., 1983, 105, 6710-6712.

Justus Leibigs Ann. Chem., 1967, 701, 206.

# COBALT CARBONYLS

86.		LECKER, S.H., NGUYEN, N.H. AND VOLLHARDT, K.P.C.	J. Am. Chem. Soc., 1986, 108, 856.
87		BUTENSCHON, H., WINKLER, M. AND VOLLHARDT, K.P.C.	J. Chem. Soc, Chem. Commun., 1986, 388-390.
88.	a.	YAMAZAKI, H. AND WAKATSUKI, Y.	Tetrahedron Lett., 1973, 3383-3384.
	b.	YAMAZAKI, H. AND WAKATSUKI, Y.	J. Chem. Soc , Chem. Commun., 1973, 280
	c.	Bonneman, H., Brinkmann, R. and Schenkluhn, H.	Synthesis, 1974, 575–577.
	d.	CLEMENT, R.A.	U S. Pat. 3829429, 1974, Dupont.
	e.	VOLLHARDT, K.P.C. AND BERGMAN, R.G.	J. Am Chem. Soc., 1974, 46, 4996-4998.
89	a.	NAIMAN, A. AND VOLLIIARDT, K.P.C.	Angew. Chem., 1977, 89, 758; Angew. Chem., Int. Edn Engl., 1977, 16, 708–709.
	b.	BRIEN, D. J., NAIMAN, A. AND Vollhardt, K.P.C.	J. Chem. Soc., Chem. Commun , 1982, 133-134.
90.		PARNELL, C.A.  and Vollhardt,  K.P.C.	Tetrahedron, 1985, 41, 5791-5796
91.		Bonneman, H.	Angew Chem., 1978, 90, 517; Angew Chem., Int Edn Engl., 1978, 17, 505–515.
92.		Hong. P. and Yamazaki, H.	Tetrahedron Lett., 1977, 1333–1336; Synthesis, 1977, 50.
<b>9</b> 3.		EARL, R.A. AND VOLLHARDT, K.P.C.	J. Am. Chem. Soc., 1983, 105, 6991-6993.
94.	a.	Schulz, A.G.	Chem Rev., 1973, 73, 385-405.
	b.	HUTCHINSON, C.R.	Tetrahedron, 1981, 37, 1047-1065.
95		Vollmardt, K.P.C.	Lect Heterocycl Chem., 1987, 9, 59.
96.		Gesing, E.R.F., Tane, J.P. and Vollhardt, K.P.C.	Angew Chem, 1980, 92, 1057; Angew Chem., Int. Edn Engl, 1980, 19, 1023-1024.
97.	a.	BURT, R., COOKE, M. AND GREEN, M.	J. Chem. Soc., 1970, 2981–2986.
	b.	CORRIGAN, P. A. AND DICKSON, R.S. FALLON, G.D., MICHEL, L.J. AND MOK, C.	Aust J Chem., 1978, 31, 1937–1951.
	c.	Corrigan, P. A. and Dickson, R. S.	Aust. J. Chem., 1979, 32, 2147-2158
	d.	Corrigan, P.A., Dickson, R.S., Johnson, S.H., Pain, G.N. and Yeoh, M.	Aust J. Chem., 1982, 35, 2203.
98.		JOHNSON, E.P. AND VOLLHARDT, K.P.C.	J. Am. Chem. Soc., 1991, 113, 381-382.
99.		GERMANAS, J., AUBERT, C. AND VOLLHARDT, K.P.C.	J. Am Chem. Soc., 1991, 113, 4006-4008.
100.		JOHNSON, B.P. AND VOLLHARDT, K.P.C.	Proc. 5th International Symposium on Natural Products Chemistry, 1992, Gordon and Breach Science.
101.		GROTJAHN, D.B. AND VOLLHARDT, K.P.C.	Synthesis, 1993, 579–605.
102.		SHEPPARD, G. AND VOLLHARDT, K.P.C.	J. Org. Chem., 1986, 51, 5496-5498.

JAVED IQBAL et al

103.	BURSE, R., KNOLKER, HJ. AND Vollhardt, K.P.C.	Angew Chem, Int Edn. Engl., 1987, 26, 1035-1037.
104	GROTJAHN, D.B. AND VOLLHARDT, K.P.C.	J. Am Chem, Soc, 1986, 108, 2091-2093.
105.	GROTIAHN, D.B AND VOLLHARDT, K.P.C.	J Am Chem. Soc., 1990, 108, 5653-5654.
106	BOESE, R., RODRIGUEZ, J. AND VOLLHARDT, K.P.C.	Angew Chem, Int Edn Engl, 1991, 30, 993-994.
107.	HARVEY, D.F., JOHNSON, B.M., UNG, C.S. AND VOLLHARDT, K.P.C	Synlett, 1989, 15-18
108	HALTERMAN, R.L. AND VOLLHARDT, K.P.C	Organometallics, 1988, 7, 883-892.
109	REVIEWS	
a	Cais, M.	Organomet Chem Rev., 1966, 1, 435.
b	HAYNES, L. AND PETTIT, R	In Carbonium ions (Olah, G.A. and Schleyer, P.V.R.,eds), Vol. 5, 1976, Wiley.
110.	SAYFERTH, D.	Adv. Organomet. Chem, 1976, 14, 97.
111.	NICHOLAS, K.M.	Acc. Chem Res , 1987, 20, 207-214
112.	NICHOLAS, K.M. AND PETTIT, R	Tetrahedron Lett, 1971, 37, 3475-3478.
113.	LOCKWOOD, R. F. AND NICHOLAS, K. M.	Tetrahedron Lett., 1978, 19, 4163-4165.
114.	HODES, H.D. AND NICHOLAS, K.M.	Tetrahedron Lett., 1978, 19, 4349-4352.
115.	SHINER, V. AND HUMPHREY, J.S.	J Am Chem. Soc., 1967, 89, 622-630.
116.	CROMBIE, L. AND MACKENZIE, K.	J Chem Soc., 1958, 4417-4435.
117.	O'BOYLE, J.E. AND NICHOLAS, K.M.	Tetrahedron Lett., 1980, 21, 1595-1598.
118.	NICHOLAS, K.M., MULVANEY, M. AND BAYER, M.	J. Am. Chem Soc, 1980, 102, 2508-2510
119.	Schreiber, S.L., Klimas, T.M. and Sammakia, T.	J. Am Chem. Soc., 1987, 109, 5149.
120.	Mikaelian, G. S., Gybin, A.S., Smit, W.A. and Caple, R.	Tetrahedron Lett., 1985, 26, 1269-1272.
121. a.	Saha, M., Bagby, B. and Nicholas, K.M.	Tetrahdron Lett, 1986, 21, 915.
ъ.	Saha, M., Muchmore, S., Van der Helm, D. and Nicholas, K.M.	J. Org. Chem, 1986, <b>51</b> , 1960.
122.	Magnus, P., Lewis, R.T. and Huffman, J.C.	J. Am Chem. Soc., 1988, 110, 6921~6923.
123.	Testev, R., Varghese, V., Montana, A.M., Khan, M. and Nicholas, K. M.	J. Org Chem, 1990, 55, 186–192.
124. a.	STUART, J. G. AND NICHOLAS, K. M.	Heterocycles, 1991, 32, 949.
b.	MONTANA, A. M. AND NICHOLAS, K. M.	J. Org. Chem., 1990, 55, 1569-1578.
125.	CAFFYN, A.J.M. AND NICHOLAS, K. M.	J. Am. Chem. Soc., 1993, 115, 6438-6439.
126. a.	Schegolev, A. V., Smit, W.A., Kalyan, Y.B., Crimer, M.Z. and Caple, R.	Tetrahedron Lett , 1982, 23, 4419-4422.

# COBALT CARBONYLS

b.	SMIT, W.A., SCHEGOLEV, A.A., Grysin, A.S., Mikalion, G.S. and Caple, R.	Synthesis, 1984, 887–890.
127.	TOP, S. AND JAOUEN, G.	J. Org. Chem., 1981, 46, 78-82.
128.	Nakagawa, M., Ma, J. and Hino, T.	Heterocycles, 1990, 30, 451-462.
129.	EL-AMOURI, H., GRUSELLE, M., JAOUEN, G., DARAN, J. C. AND VAISSERMAN, J.	J. Inorg. Chem., 1990, 29, 3238-3242.
130.	Gruselle, M., Philomin, V., Chaminant, F., Jaouen, G. and Nicholas, K.M.	J. Organomet. Chem, 1990, 399, 317-326.
131.	ROTH, KD. AND MULLER, U.	Tetrahedron Lett., 1993, 34, 2919-2922.
132.	PADMANABHAN, S. AND NICHOLAS, K.M.	J. Organomet. Chem., 1981, 212, 115-124.
133.	PADMANABHAN, S. AND NICHOLAS, K.M	Tetrahedron Lett, 1983, 24, 2239-2242.
134.	NICHOLAS, K.M. AND SIEGEL, J.	J. Am. Chem. Soc., 1985, 107, 4999-5001.
135.	STEGEL, J. AND MISLOW, K.	J. Am. Chem. Soc., 1983, 105, 7763-7764.
136.	SAHA, M. AND NICHOLAS, K.M.	J. Org. Chem, 1984, 49, 417-422.
137.	BATTISTINI, C., CROTTI, P. AND MACCHIA, F.	J. Org Chem, 1981, 46, 434-438.
138.	PADMANABHAN, S. AND NICHOLAS, K.M.	Tetrahedron Lett., 1982, 23, 2555-2558.
139. a.	Anderson, J. E.	Tetrahedron Lett, 1975, 46, 4079-4080.
ь,	JONES, E.R.H. AND MCCOMBIE, J.T.	J. Chem. Soc., 1943, 261-264.
140.	DESCOINS, C. AND SAMAIN, D.	Tetrahedron Lett., 1976, 745-748.
141.	Saha, M.	Ph. D. Dissertation, Boston College, Chestnut Hill, MA, 1985.
142.	Ojima, I.	Chem. Rev., 1988, 88, 1011-1030.
143.	Falbe, J.	New syntheses with carbon monoxide, 1994, Springer-Verlag.
144.	Bird, C.W.	Chem. Rev., 1962, 62, 283.
145. a.	HECK. R.F. AND BRESLOW, D.S.	Chem. Ind (Lond.), 1960, 467.
b.	HECK, R.F. AND BRESLOW, D.S.	J. Am. Chem. Soc., 1960, 82, 750-751.
146.	ROSENTHAL, A.	Adv. Carbohydrate Chem , 1968, 23, 59.
147. a.	HECK, R.F. AND BRESLOW, D.S.	J. Am. Chem Soc., 1960, 82, 750, 4438-4439.
b.	HECK, R. F. AND BRESLOW, D.S.	J. Am. Chem. Soc., 1961, 83, 4023.
148.	PAUSON, P.L.	InOrganometallics in organic synthesis. Aspects of a modern interdisciplinary field (de Meijere, A. and Tom Dieck, H., eds), p. 233, 1988.
149.	Guthrie, D.J.S., Khand, I.U., Knox, G.R., Kollmeier, J., Pauson, P.L. and Watts, W.E.	J. Organomet. Chem., 1975, 90, 93-100.
150. a.	Alper, H. and Currie, J.K.	Tetrahedron Lett., 1979, 2665-2666.
b.	ALPER, H., CURRIE, J.K., AND DES ABBAYES, H.	J. Chem. Soc., Chem. Cummun., 1978, 311-312.

# JAVED IQBAL et al

151	Seki, Y., Hidaka, A., Murai, S. and Sonoda, N.	Angew Chem, Int Edn Engl, 1977, 16, 174-175.
152 a.	MURAI, S. AND SONODA, N.	Angew Chem, Int. Edn Engl., 1979, 18, 837-846.
b.	CHALK, A.J. AND HARROD, J.F	J Am. Chem Soc, 1967, 89, 1640-1647.
с	BANY, Y.L. AND MACDIARMID, A.G.	Inorg Chem., 1969, 8, 986
153.	ROSENTHAL, A. AND KAN, G.	Carbohydrate Res , 1971, 19, 145.
154.	Seki, Y., Murai, S., Yamamoto, I. N. and Sonoda, N.	Angew Chem, Int Edn Engl., 1977, 789.
155.	DALCANALE, E AND FOA, M.	Synthesis, 1986, 492.
156.	EISENMANN, J. L.	J Org Chem, 1962, 27, 2706.
157.	SEKI, Y., MURAI, S. AND SONODA, N	Angew Chem, Int Edn Engl., 1978, 17, 119-120.
158.	Murai, S., Kato, T., Sonoda, N., Seki, Y. and Kawamoto, K.	Angew Chem, Int Edn Engl., 1979, 18, 393-394.
159.	Chatam, N , Murai, S. and Sonoda, N.	J. Am. Chem Soc., 1983, 105, 1370-1372.
160.	Chatani, N., Furukawa, H., Kato, T., Murai, S. and Sonoda, N.	J. Am Chem Soc., 1984, 106, 430-432.
161.	Murai, T., Kato, S., Murai, S., Toki, T., Suzuki, S. and Sonoda, N.	J Am Chem Soc., 1984, 106, 6093-6095.
162.	Chatani, N., Ikeda, T., Sano, T., Sonoda, N., Kurosawa, H., Kawasaki, Y. and Murai, S.	J Org. Chem, 1988, 53, 3387–3389.
163.	Tamag, K., Nakajo, F. and Ito, Y.	Tetrahedron Lett, 1988, 44, 39974007.
164. a.	FOA, M., FRANCALANCI, F., Bencini, E. and Gardano, A.	J. Organomet Chem, 1985, 285, 293-303
b.	Francalanci, F., Bencini, E., Gardano, A., Vincenti, M. and Foa, M.	J Organomet. Chem., 1986, 301, C27-C30.
165.	MIURA, M., AKASE, F. AND NOMURA, M.	J. Chem. Soc., Chem Commun, 1986, 241-242.
1 <b>66. a.</b>	Miura, M , Akase, F., Shinohara, M and Nomura, M.	J. Chem Soc., Perkin Trans. 1, 1987, 1021.
b.	MIURA, M., AKASE, F. AND NOMURA, M.	J. Org. Chem, 1987, 52, 2623-2625.
167.	Miura, M., Okura, K., Hattori, A. and Nomura, M.	J. Chem. Soc., Perkin Trans. 1, 1989, 73.
168.	ALPER, H. AND LAYCOCK, D.E.	Tetrahedron Lett, 1981, 22, 33-34.
169. a.	GOMBAROTTA, S. AND ALPER, H.	J. Org. Chem., 1981, 46, 2142-2144.
ь.	Alper, H.	Adv. Organomet Chem., 1981 19 183.
c.	Alper, H. and des Abbeyes, H.	J. Organomet. Chem, 1977, 134, C11.
170.	FALBE, J. AND KORTE, F.	Chem. Ber, 1965, 98, 1928-1937.
171.	ROSENTHAL, A. AND WENDER, I. H.	In Organosyntheses via metal carbonyls (Wender, I. and Pino, P., eds), 1968, Vol. 1, p.405, Interscience.
172.	Wakamatsu, H., Uda, J. and Yamakami, N.	J. Chem. Soc., Chem. Commun., 1971, 1540.

470

1.0

# COBALT CARBONYLS

- 173. SEYFERTH, D. AND SPOHN, R. J.
- 174. MURAI, T., SAKONE, T. AND KATO, S.
- 175. CHATANI, N. AND HANAFUSA, T.
- 176. TANOKA, S., TSUKIYAMA, T AND ISOBE, M.

J Am Chem. Soc., 1969, 91, 3037-3044.

- Tetrahedron Lett., 1985, 26, 5145-5148.
- J. Org. Chem, 1991, 56, 2166-2170.
- Tetrahedron Lett., 1993, 34, 5757-5760.